NOTE

Antibacterial and antilarval compounds from marine gorgonian-associated bacterium *Bacillus amyloliquefaciens* SCSIO 00856

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It is well understood that corals harbor diverse microbial communities. Recently, many coral-associated bacteria have been characterized as sources of marine natural products, especially as the coral surface is more nutrient rich than seawater or even sediments.^{1,2} Colonization of coral surfaces by bacteria and other microorganisms is mostly nondestructive to corals. Owing to the close spatial vicinity of these biofilm-forming bacteria, it can be expected that the indigenous microbial population is adapted to competitive conditions.³ The production of secondary metabolites is a common adaptation of these bacteria to compete in such microenvironments. There have been few reports documented on exploration of coralassociated bacteria in the South China Sea as a source of secondary metabolites.

In order to obtain biologically active natural products from coral-associated bacteria, we studied the marine bacterium Bacillus amyloliquefaciens SCSIO 00856 isolated from the South China Sea gorgonian Junceella juncea. In the preliminary experiments, we found that the culture broth of the strain had strong antibacterial activity towards three bacteria (Escherichia coli, Bacillus subtilis, and Staphyloccocus aureus) and antilarval activity towards laboratory-reared bryozoan Bugula neritina larvae. In this study, a new 24-membered ring lactone, macrolactin V (1) (Figure 1), together with four known compounds, macrolactin S (2),⁴ 4-butoxyphenol (3),⁵ 4-propoxyphenol (4),⁶ and 4-ethoxyphenol (5),⁶ was isolated from a culture broth of the strain. Macrolactins are polyene macrolides containing a 24-membered lactone ring. About 20 macrolactins have been chemically characterized, including macrolactins A-U, 7-O-succinyl macrolactin A, and 7-O-succinvl macrolactin F, from marine Bacillus sp. sediment isolate, and 7-O-malonvl macrolactin A from Bacillus subtilis soil isolate.^{4,7–13} Some of these compounds showed strong antibacterial, anti-tumor, and antivirus activities.^{4,7,9–13} The antibacterial activity of 1-5 against four bacterial strains and the antilarval activity of 3-5

against larval settlement of *Bugula neritina* and *Balanus amphitrite* larvae were evaluated. This paper deals with the isolation, structural elucidation of 1 and biological activity of 1–5.

The producing strain, Bacillus amyloliquefaciens SCSIO 00856, was isolated from the gorgonian coral Junceella juncea in Sanya, Hainan province, China. The strain culture (2 ml) was used to inoculate 50 ml seed medium consisting of 0.4% yeast extract, 0.4% glucose, 0.5% malt meal, and 1.8% sea salt (pH 7.3), in a 250-ml flask. A volume of 20 ml of the resulting culture was used to inoculate 400 ml of the seed medium, in a 1-l flask. The culture was incubated at 27 °C on a rotary shaker operating at 150 r.p.m. for 5 days. The resulting 400-ml culture was used to inoculate 1201 of production medium, in a 200-l fermenter, containing the following components per liter of water: yeast extract 4 g, malt meal 5 g, glucose 4 g, and sea salt 18 g. The production fermentation tanks were operated at 27 °C, a back-pressure of 0.1 MPa, and an agitation rate of 150 r.p.m. pH was controlled at 7.3 by addition of NaOH. The fermenter was operated for 10 days, at which time the culture was harvested for isolation of metabolites. The culture broth (1201) was extracted with EtOAc. The EtOAc extract was concentrated in vacuo to afford 13.9g of residue, which was subjected to column chromatography (CC) on silica gel, using CHCl₃-MeOH (from 10:0 to 0:10) as eluent. Elution of CC with 8% MeOH in CHCl₃ gave a fraction of 835 mg. The fraction was subsequently subjected to CC on silica gel, eluted with CHCl3-MeOH, then purified with semi-preparative HPLC (Luna C18(2), 250×10 mm i.d., 5 ml min⁻¹), using MeOH-water (65:35) as eluent to yield 1 (2.6 mg) and 2 (19.7 mg). Elution of the first CC with 20% MeOH in CHCl₃ gave a 342-mg fraction, which was rechromatographed on a silica gel column using a gradient elution of CHCl₃-MeOH to yield 3 (60 mg). The fraction (163 mg) eluted with 35% MeOH in CHCl₃ at the first CC was repeatedly subjected to silica gel column using a gradient elution of CHCl₃-MeOH to yield 4 (7.5 mg) and 5 (7.0 mg). The structures

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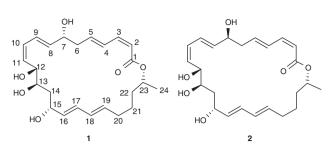


Figure 1 Structures of 1 and 2.

Table 1 $\,^{1}\text{H}$ (500 MHz) and ^{13}C (125 MHz) NMR data of 1 and 2 in CD_3OD

	1		2	
Position	δ _H , mult (J in Hz)	δ _C	δ _{H,} mult (J in Hz)	δ _C
1		167.8 C		167.8 C
2	5.58 d (12.0)	118.4 CH	5.58 d (12.0)	117.9 CH
3	6.66 dd (11.5, 11.5)	145.0 CH	6.66 dd (11.5, 11.5)	145.4 CH
4	7.25 dd (11.5, 15)	130.4CH	7.25 dd (11.5, 15)	129.9 CH
5	6.13 m	140.3 CH	6.13 m	142.1 CH
6	2.57 m	39.7 CH ₂	2.47 m	42.6 CH ₂
7	5.38 m	75.7 CH	4.18 m	73.6 CH
8	5.75 dd (7.5, 15.0)	138.5 CH	5.75 dd (7.5, 15.0)	138.8 CH
9	6.60 dd (11.5, 15.0)	130.0 CH	6.60 dd (11.5, 15.0)	127.4 CH
10	6.15 m	130.6 CH	6.15 m	130.6 CH
11	5.57 m	132.8 CH	5.57 m	131.5 CH
12	4.51 m	72.4 CH	4.51 m	72.4 CH
13	4.03 m	72.9 CH	4.03 m	73.0 CH
14	2.60 m	39.7 CH ₂	2.60 m	39.7 CH ₂
15	4.31 m	69.0 CH	4.31 m	69.0 CH
16	5.62 m	135.5 CH	5.62 m	135.5 CH
17	6.10 dd (10.5,15.0)	130.7 CH	6.10 dd (10.5, 15.0)	130.9 CH
18	5.57 dd (10.5,15.0)	131.8 CH	5.57 dd (10.5, 15.0)	131.8CH
19	5.56 m	134.8 CH	5.56 m	134.8 CH
20	2.09 m	32.7 CH ₂	2.09 m	32.7 CH ₂
	2.24 m		2.24 m	
21	1.53 m	25.5 CH_2	1.53 m	25.4 CH ₂
22	1.56 m	36.0 CH ₂	1.56 m	36.0 CH ₂
	1.65 m		1.65 m	
23	5.08 m	71.7 CH	5.08 m	71.6 CH
24	1.25 (d, 6.0)	$20.2 \ \text{CH}_3$	1.25 (d, 6.0)	20.2 CH ₃

of 3-5 were established by analyzing their 1D NMR spectral data and by comparison with the literature.^{5,6}

Compound 1 had the molecular formula of $C_{24}H_{34}O_6$ as deduced from NMR spectra and HR-ESI-MS. The ¹H NMR spectrum showed signals for one methyl group at $\delta_{\rm H}$ 1.25 (3H, d, *J*=6.0 Hz), 12 olefinic protons and five oxymethine protons at the chemical shift downfield of $\delta_{\rm H}$ 4.03–7.25 (Table 1). Its ¹³C NMR spectrum showed the presence of one methyl, five methylenes, six 1,2-disubstituted double bonds, five oxymethines, and a carbonyl group ($\delta_{\rm C}$ 167.8) (Table 1). These ¹H and ¹³C NMR data of 1 were very similar to those of macrolactin S (2)⁴ (Table 1). Extensive analyses of the ¹H NMR, ¹³C NMR, HSQC, HMBC (Figure 2), and ¹H–¹H COSY spectra of 1 inferred that 1 should have the same chemical planar structure as 2. In the ¹H NMR spectrum of 1, the ¹H coupling constants of H-2/H-3 (*J*=11.5 Hz), H-4/H-5 (*J*=15.0 Hz), H-8/H-9 (*J*=15.0 Hz), H-10/H-11 (*J*=10.0 Hz),

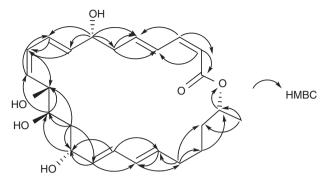


Figure 2 Key HMBC correlations of 1.

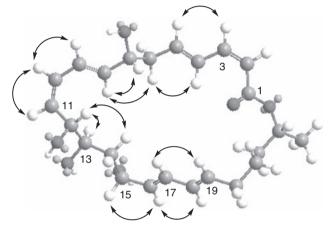


Figure 3 Key NOESY correlations of 1.

H-16/H-17 (J=15.0 Hz), and H-18/H-19 (J=15.0 Hz) indicated that geometric configurations of 1 were 2*Z*,4*E*,8*E*,10*Z*,16*E*,18*E*. The ¹H and ¹³C NMR data of H-12/C-12, H-13/C-13, H-15/C-15, and H-23/C-23 in 1 were almost the same as those in 2, which suggested that the stereochemistry of chiral carbons C-12, C-13, C-15, and C-23 was the same as that in 2.

However, compared with the ¹H and ¹³C NMR data of 2, the downfield shift of H-6 (from $\delta_{\rm H}$ 2.47 in 2 to $\delta_{\rm H}$ 2.57 in 1), H-7 (from $\delta_{\rm H}$ 4.18 to $\delta_{\rm H}$ 5.38), C-7 (from $\delta_{\rm C}$ 73.6 to $\delta_{\rm C}$ 75.7), and C-9 (from $\delta_{\rm C}$ 127.4 to $\delta_{\rm C}$ 130.0), and upfield shift of C-5 (from $\delta_{\rm C}$ 142.1 to $\delta_{\rm C}$ 140.3) and C-6 (from $\delta_{\rm C}$ 42.6 to $\delta_{\rm C}$ 39.7) were obviously observed in the NMR spectra of 1, which should be caused by the change of the three-dimensional effect at C-7 position, and suggested that the configuration of -OH at C-7 in 1 should be α -configuration instead of β -configuration as it existed in 2 and other previously isolated macrolactins.^{4,7-13} The suggestion was supported by the NOESY (Figure 3), in which NOE correlations of H-6β with H-8/H-4 and H-7 with H-8 were observed, whereas no NOE correlations between H-7 and H-9/H-5 were observed, which were in agreement with the MM2 optimized conformation of 1. However, NOE correlations of H-7 with H-9/H-5 were obviously observed in the NOESY of 2.⁴ Thus, the structure of 1 was determined as shown and named macrolactin V. The optical rotation value of 1 is -0.08 (c 0.11, MeOH), which is different from that of 2 (-0.62 (c 0.15, MeOH)).⁴

The antibacterial activity of compounds 1–5 was measured against *E. coli, Bacillus thuringiensis, Bacillus subtilis,* and *S. aureus* using standard disc diffusion assay.¹⁴ The MICs of compound 1–5 were

determined by a dilution method.¹⁵ The results showed that macrolactin V (1) had strong antibacterial activity against *E. coli, Bacillus subtilis*, and *S. aureus*, with an MIC value of $0.1 \,\mu g \, ml^{-1}$, and no activity against *Bacillus thuringiensis*, and macrolactin S (2) showed strong antibacterial activity against *E. coli* and *S. aureus* with MIC values of 0.3 and 0.1 $\mu g \, ml^{-1}$, respectively, but weak activity against *Bacillus subtilis* (MIC 100 $\mu g \, ml^{-1}$), which indicated that the configuration of OH-7 could affect the antibacterial activity of the epimers 1 and 2. Compounds 3–5 showed no or weak inhibition towards all tested bacteria.

The antilarval activity of compounds **3–5** was evaluated in settlement inhibition assays with laboratory-reared *Bugula neritina* (Bryozoa) and *Balanus amphitrite* (Cirripedia) larvae. The procedures were the same as previously reported.¹⁶ Larval settlement bioassays revealed that **3–5** showed potent antilarval activity towards *Bacillus amphitrite* larvae with EC₅₀ values of 23.0, 22.9, and 24.1 µg ml⁻¹, respectively. However, only **3** could inhibit *Bacillus neritina* larvae settlement at the concentration of 50 µg ml⁻¹, but no activity at 25 µg ml⁻¹, and **4** and **5** were inactive even at 50 µg ml⁻¹. The EC₅₀ values of **3–5** against *Bacillus amphitrite* larvae settlement were lower than the standard requirement of an EC₅₀ of 25 µg ml⁻¹ established by the US Navy program as an efficacy level for natural antifoulants, indicating that **3–5** are potential selective natural antifouling agents. This was the first time to report **3–5** from culture broths of bacteria and their antilarval activities against marine invertebrate larvae.

EXPERIMENTAL PROCEDURE

Taxonomy

Blast search results at EzTaxon.org Server showed that the 16S rRNA gene sequence of the strain SCSIO 00856 has the highest similarity (99.21%) with *Bacillus amyloliquefaciens* ATCC 23350^T. On the basis of the morphological features and the 16S rRNA gene sequence, the strain SCSIO 00856 was identified as *Bacillus amyloliquefaciens* and named *Bacillus amyloliquefaciens* SCSIO 00856. It was deposited at the Guangdong Key Laboratory of Marine Materia Medica, South China Sea Institute of Oceanology, The Chinese Academy of Sciences, China (Accession No. SCSIO 00856).

Physico-chemical properties

Macrolactin V: yellow oil; $[\alpha]_D^{20} - 0.08^{\circ}$ (c 0.11, MeOH), UV (MeOH) λ_{max} (log ε) 229 (4.57), 261 (4.18) nm; IR (KBr): 3295, 1700, 1667 cm⁻¹; for ¹H and ¹³C NMR data, see Table 1; HR-ESI-MS *m/z*: [M+H]⁺ 419.2357 (calcd. for C₂₄H₃₅O₆: 419.2355).

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