# Pentacecilides, new inhibitors of lipid droplet formation in mouse macrophages produced by Penicillium cecidicola FKI-3765-1: II. Structure elucidation 

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#### Abstract

The structures of pentacecilides, new inhibitors of lipid droplet formation in mouse macrophages produced by Penicillium cecidicola FKI-3765-1, were elucidated by spectroscopic studies, including various NMR experiments. Pentacecilides have a common pentacyclic meroterpene core, which contains an aromatic ring and a $\delta$-lactone ring.


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## INTRODUCTION

Three new compounds, designated pentacecilides A to C (Figure 1) were isolated as inhibitors of lipid droplet formation in mouse macrophages from the culture broth of $P$. cecidicola FKI-3765-1. ${ }^{1}$ The taxonomy of the producing strain, fermentation, isolation and biological properties of pentacecilides were described in an earlier paper. ${ }^{1}$ In this study, the physicochemical properties and structure elucidation of pentacecilides are described.

## RESULTS

## Physicochemical properties

The physicochemical properties of pentacecilides A to C are summarized in Table 1. They have a similar pattern with absorption maxima at $214-219 \mathrm{~nm}, 273-274 \mathrm{~nm}$ and $309-310 \mathrm{~nm}$ in UV spectra. IR absorption at $1619-1745 \mathrm{~cm}^{-1}$ and $3401-3434 \mathrm{~cm}^{-1}$ suggested the presence of carbonyl and hydroxy groups in their structures. These data indicated that they share a similar skeleton.

## Structure elucidation of pentacecilide C

The molecular formula of pentacecilide C was determined to be $\mathrm{C}_{27} \mathrm{H}_{34} \mathrm{O}_{8}$ on the basis of HRESI-TOF-MS measurement (Table 1). The ${ }^{13} \mathrm{C}$ NMR spectrum (in $\mathrm{CDCl}_{3}$ ) showed 27 resolved signals, which were classified into six methyl carbons, four methylene carbons, two $s p^{3}$ methine carbons, one $s p^{2}$ methine carbon, three oxygenated $s p^{3}$ methine carbons, two $s p^{3}$ quaternary carbons, one oxygenated $s p^{3}$ quaternary carbon, three $s p^{2}$ quaternary carbons, two oxygenated $s p^{2}$ quaternary carbons and three carbonyl carbons by analysis of the DEPT and heteronuclear single quantum coherence (HSQC) spectra.

The ${ }^{1} \mathrm{H}$ NMR spectrum ( $\mathrm{in} \mathrm{CDCl}_{3}$ ) displayed 33 proton signals, one of which was suggested to be a hydroxyl proton ( $\delta 11.08$ ), as reported in thailandolides. ${ }^{2}$ Taking the molecular formula into consideration, the presence of another hydroxy proton was suggested. The connectivity of proton and carbon atoms was established by the ${ }^{13} \mathrm{C}-{ }^{1} \mathrm{H}$ HSQC spectrum (Table 2). Analyses of ${ }^{1} \mathrm{H}-{ }^{1} \mathrm{H}$ COSY revealed the presence of partial structures I to IV, as shown in Figure 2. Furthermore, ${ }^{13} \mathrm{C}-{ }^{1} \mathrm{H}$ long-range couplings of ${ }^{2} J$ and ${ }^{3} J$ observed in the ${ }^{13} \mathrm{C}-{ }^{1} \mathrm{H}$ HMBC spectrum gave the following linkages (Figure 3a): (1) Cross-peaks from $\mathrm{H}_{2}-7^{\prime}(\delta 2.70,2.85)$ to $\mathrm{C}-1^{\prime}(\delta 110.7), \mathrm{C}-2^{\prime}(\delta 139.3)$ and $\mathrm{C}-3^{\prime}$ ( $\delta$ 102.2), from $\mathrm{OH}-4^{\prime}(\delta 11.08)$ to $\mathrm{C}-3^{\prime}, \mathrm{C}-4^{\prime}(\delta 162.3)$ and $\mathrm{C}-5^{\prime}$ ( $\delta 103.5$ ) and from $\mathrm{H}-5^{\prime}(\delta 6.30)$ to $\mathrm{C}-1^{\prime}, \mathrm{C}^{\prime}-3^{\prime}, \mathrm{C}-4^{\prime}$ and $\mathrm{C}-6^{\prime}(\delta 159.1)$ indicated that a phenol skeleton connects the partial structure I at $\mathrm{C}-2^{\prime}$. Furthermore, the findings that the chemical shift of $\mathrm{C}-8^{\prime}(\delta 74.7)$ corresponds to an oxygenated carbon and the $\mathrm{OH}-4^{\prime}$ proton ( $\delta$ 11.08) shifted to a lower field because of a hydrogen bonding indicated that $\mathrm{C}-3^{\prime}$ and $\mathrm{C}-8^{\prime}$ are connected through an ester bond, which form $\delta$-lactone. This was also supported by the IR absorption (1619$1666 \mathrm{~cm}^{-1}$ ). Although observation of a cross-peak from H-8' to $\mathrm{C}-10^{\prime}(\delta 169.9)$ was important and simple to show the presence of $\delta$-lactone, the cross-peak was not observed because the dihedral angle between $\mathrm{H}-8^{\prime}$ and $\mathrm{C}-10^{\prime}$ is $90^{\circ}$. Therefore, the coupling constant in HMBC experiment was changed from $J_{\mathrm{C}-\mathrm{H}}=8.0 \mathrm{~Hz}$ to $J_{\mathrm{C}-\mathrm{H}}=3.0 \mathrm{~Hz}$. As a result, the long-range coupling of ${ }^{4} \mathrm{~J}$ from $\mathrm{H}-5^{\prime}$ to $\mathrm{C}-10^{\prime}$ was observed, supporting the presence of $\delta$-lactone. (2) Cross-peaks from $\mathrm{H}_{2}-11(\delta 2.52)$ to $\mathrm{C}-8(\delta 79.0)$, from $\mathrm{H}_{3}-12(\delta 1.23)$ to $\mathrm{C}-7$ ( $\delta 71.8$ ), C-8 and C-9 ( $\delta 43.0$ ), from H-7 ( $\delta 4.11$ ) to C-8 and C-9, from H-9 ( $\delta 2.23$ ) to C-8, C-10 ( $\delta 36.2$ ) and C-15 ( $\delta 24.6$ ), from $\mathrm{H}_{2}-6$

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Thailandolide A


Thailandolide B

Figure 1 Structures of pentacecilides $A$ to $C$ and thailandolides $A$ and $B$.

Table 1 Physicochemical properties of pentacecilides A, B and C

|  | Pentacecilide A | Pentacecilide B | Pentacecilide C |
| :---: | :---: | :---: | :---: |
| Appearance | White crystalline solid | White crystalline solid | White crystalline solid |
| Molecular weight | 412 | 470 | 486 |
| Molecular formula | $\mathrm{C}_{25} \mathrm{H}_{32} \mathrm{O}_{5}$ | $\mathrm{C}_{27} \mathrm{H}_{34} \mathrm{O}_{7}$ | $\mathrm{C}_{27} \mathrm{H}_{34} \mathrm{O}_{8}$ |
| HRESI-TOF-MS ( $\mathrm{m} / \mathrm{z}$ ) |  |  |  |
| Calcd: | $435.2147(\mathrm{M}+\mathrm{Na})^{+}$ | $493.2225(\mathrm{M}+\mathrm{Na})^{+}$ | $509.2151(\mathrm{M}+\mathrm{Na})^{+}$ |
| Found: | $435.2141(\mathrm{M}+\mathrm{Na})^{+}$ | $493.2202(\mathrm{M}+\mathrm{Na})^{+}$ | $501.2162(\mathrm{M}+\mathrm{Na})^{+}$ |
| UV ( MeOH ) $\lambda_{\text {max }} \mathrm{nm}(\varepsilon)$ | 219 (18647), 274 (11846), 309 (4194) | 219 (33055), 274 (18226), 309 (4784) | 214 (54432), 273 (32736), 310 (4947) |
| $[\alpha]]^{6}$ | $-4.38{ }^{\circ}\left(c=0.38, \mathrm{CHCl}_{3}\right)$ | $-32.6^{\circ}\left(\mathrm{c}=0.68, \mathrm{CHCl}_{3}\right)$ | $-32.3^{\circ}\left(\mathrm{c}=0.48, \mathrm{CHCl}_{3}\right)$ |
| $\mathrm{IR}(\mathrm{KBr}) v_{\text {max }}\left(\mathrm{cm}^{-1}\right)$ | 3401, 1697. 1662, 1465, 1380 | 3440, 1747, 1727, 1666, 1475 | 3434, 1745, 1666, 1619, 1473 |

$(\delta 1.86,2.18)$ to $\mathrm{C}-10$, from $\mathrm{H}_{3}-15(\delta 1.54)$ to $\mathrm{C}-9, \mathrm{C}-10$ and $\mathrm{C}-1(\delta$ 41.1), from H-5 $(\delta 1.82)$ to $\mathrm{C}-4(\delta 48.3), \mathrm{C}-10$ and $\mathrm{C}-15$, from $\mathrm{H}_{2}-1(\delta$ $1.88,2.20)$ to $\mathrm{C}-3(\delta 207.8), \mathrm{C}-5, \mathrm{C}-10$ and $\mathrm{C}-15$, from $\mathrm{H}_{3}-13(\delta 1.14)$ to $\mathrm{C}-3, \mathrm{C}-4, \mathrm{C}-5$ and $\mathrm{C}-14(\delta 25.7)$, from $\mathrm{H}_{3}-14(\delta 1.20)$ to $\mathrm{C}-3, \mathrm{C}-4$, $\mathrm{C}-5$ and $\mathrm{C}-13(\delta 20.8)$ and from $\mathrm{H}-2(\delta 5.63)$ to $\mathrm{C}-3$ showed the presence of a 3-oxo-decalin skeleton containing the partial structures II to IV. (3) Cross-peaks from $\mathrm{H}-2$ and $\mathrm{H}_{3}-17$ ( $\delta 2.17$ ) to C-16 ( $\delta$ 170.3) showed that an acetoxy group is connected to C-2. The chemical shift of C-7 ( $\delta 71.8$ ) and the molecular formula showed the presence of a hydroxy group. (4) The finding that cross-peaks were
observed from $\mathrm{H}-11$ to $\mathrm{C}-1^{\prime}, \mathrm{C}-2^{\prime}$ and $\mathrm{C}-6^{\prime}$ and that the chemical shifts of $\mathrm{C}-8(\delta 79.0)$ and $\mathrm{C}-6^{\prime}(\delta 159.1)$ correspond to an oxygenated carbon indicated that a phenol and a decalin ring are connected by a pyran ring. The pentacyclic structure was found to consist of a sixmembered lactone, a phenol, a pyran and a decalin ring. Thus, the structure of pentacecilide $C$ was elucidated as shown in Figure 1. The structure satisfied the degree of unsaturation and the molecular formula. Furthermore, all chemical shifts, except for C-2 in pentacecilide $C$, were comparable with those reported for thailandolide A. ${ }^{2}$

Table $2{ }^{1} \mathrm{H}$ and ${ }^{13} \mathrm{C}$ NMR chemical shift of pentacecilides $\mathrm{A}, \mathrm{B}$ and C

| No. | Pentacecilide A |  | Pentacecilide B |  | Pentacecilide C |  |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: |
|  | $\delta_{C}$ | $\delta_{H}(\mathrm{~J}$ in Hz) | $\delta_{C}$ | $\delta_{H}(\mathrm{~J}$ in Hz) | $\delta_{C}$ | $\delta_{H}(\mathrm{~J}$ in Hz) |
| 1 | 31.7 | 1.67 m | 40.7 | 1.85 m | 41.1 | 1.88 m |
|  |  | 2.06 m |  | 2.18 m |  | 2.20 m |
| 2 | 33.7 | 2.44 m | 72.6 | 5.59 (12.0, 7.0) | 72.2 | 5.63 (12.0, 7.0) |
|  |  | 2.68 m |  |  |  |  |
| 3 | 219.6 | - | 208.6 | - | 207.8 | - |
| 4 | 47.2 | - | 48.6 | - | 48.3 | - |
| 5 | 48.1 | 1.88 m | 48.0 | 1.77 m | 45.6 | 1.82 m |
| 6 | 18.5 | 1.58 m | 17.7 | 1.68 m | 26.7 | 1.86 m |
|  |  |  |  | 1.75 m |  | 2.18 m |
| 7 | 33.7 | 2.08 m | 36.6 | 2.10 m | 71.8 | 4.11 dd (10.0, 3.0) |
|  |  | 2.24 m |  |  |  |  |
| 8 | 78.3 | - | 78.1 | - | 79.0 | - |
| 9 | 44.8 | 1.86 m | 48.5 | 1.88 m | 43.0 | 2.23 m |
| 10 | 35.4 | - | 36.6 | - | 36.2 | - |
| 11 | 20.3 | 2.44 m | 21.2 | 2.44 dd (10.0, 2.5) | 20.7 | 2.52 m |
| 12 | 25.5 | 1.35 s | 24.0 | 1.26 s | 21.2 | 1.23 s |
| 13 | 20.0 | 1.09 s | 21.3 | 1.18 s | 20.8 | 1.20 s |
| 14 | 29.3 | 1.12 s | 26.3 | 1.16 s | 25.7 | 1.14 s |
| 15 | 23.0 | 0.96 s | 24.9 | 1.39 s | 24.6 | 1.54 s |
| 16 |  |  | 170.3 | - | 170.3 | - |
| 17 |  |  | 21.0 | 2.17 s | 21.0 | 2.17 s |
| $1^{\prime}$ | 110.5 | - | 110.9 | - | 110.7 | - |
| $2^{\prime}$ | 139.2 | - | 139.3 | - | 139.3 | - |
| $3 \prime$ | 101.7 | - | 101.9 | - | 102.2 | - |
| $4^{\prime}$ | 162.5 | - | 162.6 | - | 162.3 | - |
| $5^{\prime}$ | 103.3 | 6.27 s | 103.5 | 6.27 s | 103.5 | 6.30 s |
| $6^{\prime}$ | 160.4 | - | 160.2 | - | 159.1 | - |
| 7 | 32.0 | $2.72 \mathrm{dd}(17.0,11.0)$ | 32.1 | $2.70 \mathrm{dd}(17.0,11.0)$ | 31.8 | $2.70 \mathrm{dd}(17.0,11.0)$ |
|  |  | 2.84 dd (17.0, 3.5) |  | 2.85 dd (17.0, 3.5) |  | 2.85 dd (17.0, 3.5) |
| $8^{\prime}$ | 75.0 | 4.64 m | 74.9 | 4.62 m | 74.7 | 4.62 m |
| $9^{\prime}$ | 21.2 | 1.54 d (7.0) | 21.2 | 1.55 d (7.0) | 20.9 | 1.55 d (7.0) |
| $10^{\prime}$ | 170.5 | - | 170.3 | - | 169.9 | - |
| 4'-OH |  | 11.08 s |  | 11.07 s |  | 11.08 s |



I


III


II


IV

Figure 2 Partial structures of pentacecilide C.

## Structure elucidation of pentacecilide B

The molecular formula $\mathrm{C}_{27} \mathrm{H}_{34} \mathrm{O}_{7}$ of pentacecilide B is smaller by one oxygen atom than that of pentacecilide C . Comparison of the ${ }^{1} \mathrm{H}$ NMR spectra between pentacecilides $B$ and $C$ indicated that the oxygenated $s p^{3}$ methine proton (H-7) in pentacecilide C is replaced by methylene protons ( $\delta 2.10$ ) in pentacecilide B. In fact, analyses of the ${ }^{1} \mathrm{H}-{ }^{1} \mathrm{H}$ COSY revealed the presence of the partial structure V containing the replaced part (Figure 3b). The partial structure V was also confirmed by observing cross-peaks from $\mathrm{H}_{3}-12(\delta 1.26)$ to C-7 ( $\delta 36.6$ ), $\mathrm{C}-8(\delta 78.1)$ and $\mathrm{C}-9(\delta 48.5)$, from $\mathrm{H}_{2}-7(\delta 2.10)$ to $\mathrm{C}-8$, from H-9 $(\delta 1.88)$ to C-8, C-10 $(\delta 36.6)$ and C-15 ( $\delta 24.9$ ), from $\mathrm{H}_{2}-6$ $(\delta 1.68,1.75)$ to $\mathrm{C}-10$, from $\mathrm{H}_{3}-15(\delta 1.39)$ to $\mathrm{C}-9$ and $\mathrm{C}-10$ and from $\mathrm{H}-5(\delta 1.77)$ to $\mathrm{C}-10$ in ${ }^{13} \mathrm{C}-{ }^{1} \mathrm{H}$ HMBC experiments (Figure 3b). Taken together, the structure of pentacecilide B was elucidated as 7-dehydroxy pentacecilide C (Figure 1).

## Structure elucidation of pentacecilide A

The molecular formula $\mathrm{C}_{25} \mathrm{H}_{32} \mathrm{O}_{5}$ of pentacecilide A is smaller by $\mathrm{C}_{2} \mathrm{H}_{2} \mathrm{O}_{3}$ than that of pentacecilide C. Comparison of the ${ }^{1} \mathrm{H}$ NMR spectra of pentacecilides A and B showed that the methyl protons $\left(\mathrm{H}_{3}-17\right)$ disappear and the oxygenated $s p^{3}$ methine proton (H-2) is


Figure $3^{1} \mathrm{H}-{ }^{1} \mathrm{H}$ COSY and ${ }^{13} \mathrm{C}-{ }^{1} \mathrm{H}$ HMBC experiments of pentacecilides $\mathrm{A}(\mathbf{a}), \mathrm{B}(\mathrm{b})$ and $\mathrm{C}(\mathbf{c})$.
replaced by methylene protons ( $\delta 2.44,2.68$ ). Analyses of the ${ }^{1} \mathrm{H}-{ }^{1} \mathrm{H}$ COSY revealed that the partial structure VI contains the replaced part (Figure 3c). The partial structure VI was also confirmed by observing cross-peaks from $\mathrm{H}_{3}-15(\delta 0.96)$ to $\mathrm{C}-1(\delta 31.7)$ and C-10 ( $\delta 35.4$ ), from $\mathrm{H}-5(\delta 1.88)$ to $\mathrm{C}-4(\delta 47.2)$ and $\mathrm{C}-10$, from $\mathrm{H}_{2}-1(\delta 1.67,2.06)$ to $\mathrm{C}-3(\delta 219.6), \mathrm{C}-5(\delta 48.1), \mathrm{C}-10$ and $\mathrm{C}-15(\delta 23.0)$, from $\mathrm{H}_{3}-13(\delta$ 1.09) to C-3, C-4, C-5 and C-14 ( $\delta 29.3$ ), from $\mathrm{H}_{3}-14(\delta 1.12)$ to $\mathrm{C}-3$, C-4, C-5 and C-13 (20.0) and from $\mathrm{H}_{2}-2$ to $\mathrm{C}-3$ in ${ }^{13} \mathrm{C}-{ }^{1} \mathrm{H}$ HMBC experiments (Figure 3c). Taken together, the structure of pentacecilide A was elucidated as 2-deacetoxy pentacecilide B (Figure 1).

## Stereochemistry of pentacecilides

Pentacecilide C has seven chiral carbons in its structure. The relative stereochemistry at C-2, C-5, C-7, C-8, C-9 and C-10 of the tricyclic skeleton (A-B-C) was investigated by NOESY experiments. As shown in Figure 4, cross-peaks were observed between $\mathrm{H}-2(\delta 5.63)$ and $\mathrm{H}_{3}-$ $13(\delta 1.20) / \mathrm{H}_{3}-15(\delta 1.54)$, between $\mathrm{H}_{3}-15$ and $\mathrm{H}_{a x}-6(\delta 2.18) / \mathrm{H}_{3}-13$, between H-5 ( $\delta 1.82$ ) and H-7 ( $\delta 4.11$ )/H-9 ( $\delta 2.23$ ) and H-7 and $\mathrm{H}-9$, indicating that they are oriented in a 1,3-diaxial conformation. Accordingly, rings A and B are oriented in a chair-chair form. Secondly, cross-peaks were observed between $\mathrm{H}_{3}-12(\delta 1.23)$ and $\mathrm{H}-7 / \mathrm{H}_{2}-11(\delta 2.52)$ and between $\mathrm{H}_{3}-15$ and $\mathrm{H}_{2}-11$, suggesting that rings B and C are oriented in a chair-boat form. Thirdly, cross-peaks were observed between $\mathrm{H}_{e q-}{ }^{-7^{\prime}}(\delta 2.85)$ and $\mathrm{H}-11 / \mathrm{H}-8^{\prime}(\delta 4.62)$ and


Figure 4 NOESY experiments of pentacecilide C.
between $\mathrm{H}_{a x}-7^{\prime}$ and $\mathrm{H}_{3}-9(\delta 1.55)$, indicating that the conformation of $\mathrm{H}_{3}-9^{\prime}$ is equatorial, which was also supported by a large coupling constant ( $J=11 \mathrm{~Hz}$ ) between $\mathrm{H}_{a x}-7^{\prime}$ and $\mathrm{H}-8^{\prime}$. Taken together, the relative stereochemistry of pentacecilide C was determined as shown in Figure 1. These results were consistent with those of thailandolide A except for the stereochemistry at $\mathrm{C}-8^{\prime}$, although chemical shifts at $\mathrm{C}-8^{\prime}$ of pentacecilide C and thailandolide A were almost the same. ${ }^{2}$

The relative stereochemistry of pentacecilides A and B was deduced to be the same as that of pentacecilide C by the similarity of NOESY experiments and the coupling constants in ${ }^{1} \mathrm{H}$ NMR.

## DISCUSSION

Pentacecilides A to C, structurally related to thailandolides, were isolated from the culture broth of P. cecidicola FKI-3765-1, and were found to have a common pentacyclic core containing an aromatic ring and a $\delta$-lactone ring. The core seemed to be a meroterpene, consisting of a sesquiterpene and a pentaketide. Thailandolides were reported to be produced by Talaromyces thailandiasis, ${ }^{2}$ whereas pentacecilides were produced by a different genus Penicillium. Thailandolides A and B were not detected in the culture broth of P. cecidicola FKI-3765-1.

From the structure elucidation, the planar structures of pentacecilides were seen to be similar to those of thailandolides. The relative stereochemistry of the compounds is almost the same, but the C-8' stereochemistry is different; the $\mathrm{C}-8^{\prime}$ methyl group of thailandolides is
oriented in the axial conformation, ${ }^{2}$ whereas that of pentacecilides is in the equatorial conformation.

## METHODS

## General experimental procedures

UV spectra were recorded on a spectrophotometer ( 8453 UV-Visible spectrophotometer, Agilent Technologies Inc., Santa Clara, CA, USA). IR spectra were recorded on a Fourier transform infrared spectrometer (FT-710, Horiba Ltd, Kyoto, Japan). Optical rotations were measured with a digital polarimeter (DIP-1000, JASCO Corporation, Tokyo, Japan). ESI-TOF-MS and HRESI-TOF-MS spectra were recorded on a mass spectrometer (JMS-T100LP, JEOL Ltd, Tokyo, Japan). Various NMR spectra were measured with a spectrometer (XL-400, Varian Inc., Palo Alto, CA, USA).

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