NOTE



JBIR-26, a Novel Natural Compound from *Streptomyces* sp. AK-AH76, Regulates Mammalian Circadian Clock

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Abstract In the course of our screening program for regulators of circadian clock system (circadian rhythms), we isolated a new natural compound JBIR-26 (1) from *Streptomyces* sp. AK-AH76. The structure was determined to be 2-hydroxy-3,6-dimethylbenzoic acid on the basis of spectroscopic data. Compound 1 lengthened the periodlength of mammalian clocks for 1.5 hours at a concentration of $10 \, \mu \text{g/ml}$.

Keywords 2-hydroxy-3,6-dimethylbenzoic acid, JBIR-26, *Streptomyces* sp., circadian clock, *PER2*

Daily rhythms play a prominent role in everything from sleep/wake cycles, body temperature, hormone levels, and even cognition, attention and mood. Disruptions in circadian rhythms associate with not only abnormal sleep/wake but also the pathogenesis of mood disorders such as bipolar disorder, major depression and seasonal affective disorder [1]. Therefore, controlling the clock systems to maintain accurate circadian rhythms will be one of therapeutic strategies for those disorders. However, because the mammalian circadian clock consists of complexly integrated transcriptional feed-back and feed-forward loops of a set of clock genes and regulated protein

turnover to maintain 24-hour periodicity $[2\sim5]$, it would be difficult to logically derive such a control method. In this situation, chemical screening using high-throughput assays is a powerful approach [6].

In mammals, the circadian clock system is composed of a central pacemaker in the suprachiasmatic nucleus and subsidiary oscillators in most peripheral tissues. Recent studies have also shown that even cultured fibroblasts contain a self-sustainable circadian clock [7]. A high-throughput method to monitor the dynamics of the clock of these fibroblasts has been developed [8]. Therefore, these fibroblasts are useful tool to verify the influence of perturbations on mammalian circadian clock [9 \sim 11]. In the course of our screening program for circadian clock regulators with a clock-controlled reporter system, we isolated a novel natural product designated as JBIR-26 (1) from *Streptomyces* sp. AK-AH76 (Fig. 1). We report herein

1: 2-hydroxy-3,6-dimethylbenzoic acid

Fig. 1 Structure of JBIR-26 (1).

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Table 1 Physico-chemical properties of JBIR-26 (1)

Appearance	Colorless amorphous solid
Melting point ^a	98~101°C
Molecular formula	$C_9H_{10}O_3$
HR-ESI-MS (<i>m/z</i>) ^b	found: 165.0555 [M-H] ⁻
	calcd: 165.0552
UV $\lambda_{ m max}$ (MeOH) nm ($arepsilon$) $^{ m c}$	311 (4,070), 249 (5,550)
IR $v_{\sf max}$ (KBr) cm $^{-1}$ d	1646, 1247

^a Melting point was determined with a Yanagimoto micro melting point apparatus. ^b HR-ESI-MS data were recorded on a Waters LCT-Premier XE. ^c UV spectrum was measured on a BECKMAN COULTER DU730 UV/Vis spectrophotometer. ^d IR spectrum was obtained using a HORIBA FT-720 spectrophotometer.

the fermentation, isolation, structure determination and brief biological activity of 1.

Streptomyces sp. AK-AH76 isolated from soil sample collected in Tokyo, Japan, was cultured at 27°C for 5 days in 500-ml Erlenmeyer flasks containing 100 ml of a medium consisting of 2.5% starch, 1.5% soybean meal, 0.2% dry yeast and 0.4% CaCO₃ (pH 6.2 before sterilization) on a rotary shaker (180 rpm). The fermentation broth (2.0 liters) was centrifuged and the supernatant was extracted with EtOAc and evaporated to dryness. The organic layer (320 mg) was chromatographed on a silica gel flash column (Purif-Pack SI-60, Moritex) with a CHCl₃-MeOH linear gradient system ($0\sim100\%$ MeOH), and fractions including active metabolites were collected. The 10~50% eluate (117 mg) was further purified by ODS flash column (Purif-Pack ODS, Moritex) with 40% aqueous MeOH to give an active fraction (16 mg). The fraction was further purified by reversed-phase preparative HPLC using PEGASIL ODS (Senshu Pak, 20 i.d.×150 mm) developed with 40% aqueous MeOH with 0.2% formic acid to yield JBIR-26 (1, Rt=26.0 minutes, 1.8 mg).

The physico-chemical properties of 1 are summarized in Table 1. Compound 1 was isolated as a colorless amorphous solid, and the molecular formula of 1 was established as $C_9H_{10}O_3$ on the basis of HR-ESI-MS analysis (found: $165.0555 \ [M-H]^-$, calcd 165.0552). The ^{13}C - and ^{1}H -NMR spectral data for 1 are shown in Table 2. The direct connectivity of protons and carbons was established by the HSQC spectrum.

The structure of **1** was elucidated by NMR spectral analysis such as ^{1}H - ^{1}H DQF-COSY and HMBC spectra. The ^{1}H - ^{1}H ortho-coupling between 4-H ($\delta_{\rm H}$ 7.11) and 5-H ($\delta_{\rm H}$ 6.60) indicated the presence of a 1,2,3,4-tetrasubstituted benzene ring moiety. ^{1}H - ^{13}C m-couplings in the HMBC spectrum from 4-H to aromatic quaternary

Table 2 ¹³C- (150 MHz) and ¹H- (600 MHz) NMR data for **1**

Position	$\delta_{ extsf{C}}$	δ_{H}
1	113.5	
1-COOH	175.7	
2	162.4	
3	124.7	
3-Me	15.9	2.15 (s, 3H)
4	135.5	7.11 (d, 7.8)
5	122.8	6.60 (d, 7.8)
6	140.2	
6-Me	23.9	2.51 (s, 3H)

NMR spectra were recorded on a Varian NMR System 600 NB CL. The sample was measured in CHD₂OD, and the solvent peak was used as an internal standard ($\delta_{\rm H}$ 3.30 and $\delta_{\rm C}$ 49.0).

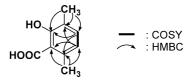


Fig. 2 Key correlations in 2D NMR of 1.

carbons C-2 ($\delta_{\rm C}$ 162.4) and C-6 ($\delta_{\rm C}$ 140.2), and from 5-H to aromatic quaternary carbons C-1 ($\delta_{\rm C}$ 113.5) and C-3 ($\delta_{\rm C}$ 124.7) revealed assignments of these carbons. The ¹H-¹³C long-range couplings were observed from singlet methyl group 3-Me ($\delta_{\rm H}$ 2.15) to C-2, C-3, and C-4 ($\delta_{\rm C}$ 135.5), and another singlet methyl group 6-Me ($\delta_{\rm H}$ 2.51) to C-1, C-5 ($\delta_{\rm C}$ 122.8), and C-6. The remaining units were a carbonyl carbon which should be assigned to carboxylic carbonyl carbon and a hydroxyl group from the molecular formula. By taking into consideration the ¹³C chemical shifts of C-1 and C-2, the carboxylic acid and the hydroxyl residue should be substituted at the positions of C-1 and C-2, respectively. Therefore, the structure of 1 was determined as 2-hydroxy-3,6-dimethylbenzoic acid (Fig. 1). Although this compound has been already reported as a degradation product of polyketomycin from Streptomyces sp. MK277-AF1 [12], this is the first report of its isolation as a natural product.

To evaluate the effect on the state of circadian clocks by 1, NIH3T3 cells were stably transfected with luciferase gene driven by the promoter of PER2 clock gene and stimulated with forskolin to synchronize circadian rhythmicity [9 \sim 11]. After addition of 1, we monitored multicell-level PER2-luciferase reporter activity in real time. The period-length of oscillation of PER2-luciferase

reporter activity in non-treated cells was 20.8 ± 0.3 hours, but the period-length was lengthened to 22.3 ± 0.8 hours by treatment with $10\,\mu\text{g/ml}$ of 1. Thus, 1 exhibited a 1.5 hours period-lengthening in transcriptional cycles of clock gene reporter compared with the control. Studies on detailed biological activity of 1 are now underway.

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