brief communications

Pal et al. used a data set of 4,273 deletion effects, each of which is based on two replicate measurements; we used a new data set of 5,937 deletion effects, 4,744 of which are based on more than 10 replicate measurements (data by courtesy of J. Kumm and G. Giaever). Calculating Kendall's correlation coefficient⁸ between deletion effect and evolutionary distance, we observed a highly significant negative relationship $(-0.26 < \tau < -0.17, P < 0.0001$ in all evolutionary comparisons; manuscript in preparation). To control for different levels of gene expression, we used recent expression data9 that were not measured in an aneuploid strain of yeast¹⁰. Calculating Kendall's partial correlation coefficient⁸ $(-0.18 < \tau < -0.16, P < 0.0001$ in all evolutionary comparisons), we find that the relationship between protein dispensability and evolutionary rate remains highly significant, even when controlling for geneexpression levels.

Aaron E. Hirsh*, Hunter B. Fraser†

*Department of Biological Sciences, Stanford University, Stanford, California 94305, USA e-mail: aehirsh@stanford.edu †Department of Molecular and Cell Biology,

University of California, Berkeley,

California 94720, USA

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COMMUNICATIONS ARISING Cosmology

Do black holes constrain varying constants?

entative observations of shifted spectral lines in distant quasars^{1,2} have rekindled interest in Dirac's old idea^{3,4} that the fundamental 'constants' of physics may vary over time. Davies et al. have argued that black-hole thermodynamics favours theories in which the speed of light, c, decreases, and does not favour those in which the fundamental electronic charge, e, increases⁵. Here we show, however, that when the entire thermal environment of a black hole is considered, no such conclusion can be drawn. Although black-hole features such as mass quantization may still constrain models with varying 'constants'6, thermodynamics probably cannot.



The observations of refs 1, 2 suggest that the fine-structure constant, α , may have been slightly smaller in the early Universe. As $\alpha = e^2/\hbar c$ depends on *e*, *c* and Planck's constant, \hbar , it is natural to ask which of these values varies. Davies et al. offer an ingenious argument. A black hole with mass M and charge Q = ne has an entropy⁷ $S = \pi G/\hbar c [M + (M^2 - n^2 e^2/G)^{1/2}]^2$. Evidently, an increase in e will reduce the entropy, violating the generalized second law of thermodynamics, whereas a decrease in \hbar or c will increase the entropy. This argument involves assumptions that may not be valid for all models^{6,8,9}, but it offers an interesting starting point.

As Davies *et al.* note, however, such an argument should consider not just the black hole, but also its surroundings. An isolated black hole is never in thermal equilibrium: it decays by Hawking radiation and, if it is charged, by spontaneous emission of charged particles¹⁰. These processes reduce *S*, but do not violate the second law of thermodynamics because there is a compensating increase in the entropy of the environment.

To investigate the thermodynamics of varying 'constants', one should study a black hole that is in equilibrium with its environment. This can be done by considering a black hole in a 'box' of radius $r_{\rm B}$, with fixed boundary temperature *T* and charge *Q* (the canonical ensemble) or electrostatic potential ϕ (the grand canonical ensemble). Note that $r_{\rm B}$ can be altered only by doing work on the system.

In the canonical ensemble, the entropy is given by $S = \pi r_B^2 x^2$, where *x* is determined by the seventh-order equation¹¹ $x^5(x-q^2)(x-1) + b^2(x^2-q^2)^2 = 0$, where $q = \sqrt{GQ/r_Bc^2}$ and $b = \hbar c/4\pi r_B kT$. Figure 1 shows a plot of S/r_B^2 against q^2 and *b*. It is apparent — and may be confirmed numerically — that the entropy increases with increasing α . For the grand canonical ensemble, exact analytical results lead to the same conclusion. Black-hole thermodynamics thus militates against models in which the fundamental charge, *e*, decreases, but places no restriction on increasing *e*.

To compare this result with that of Davies *et al.*, note first that the Hawking temperature of a charged black hole decreases with increasing *e*. A black hole will thus cool below the ambient temperature of the heat bath and will absorb heat, thereby increasing its mass. According to the first law of thermodynamics, the net change in entropy is $dS = 1/T(dE - \phi dQ)$, and it may be verified that the increase in the energy, *E*, dominates.

Of course, such thermodynamic arguments only describe relationships among equilibria, and not the transitions between equilibria. A more detailed analysis, however, requires an explicit, dynamic model. In particular, any theory with a variable fine-structure constant necessarily contains a new scalar field, α itself, the entropy of which cannot be neglected during dynamic processes in which α is changing. Jacobson (personal communication) has suggested that a suitable dynamic version of the second law of thermodynamics¹² will ensure that entropy increases during such a process. Black-hole thermodynamics is therefore insufficient to constrain theories in which α increases.

Steven Carlip, Sachindeo Vaidya

Department of Physics, University of California, Davis, California 95616, USA

e-mail: carlip@dirac.ucdavis.edu

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