

violet. It forms a picrate m.p. 117.5° (found: C, 35.9; H, 2.8; N, 24.6. $C_{12}H_7O_7N_7 \cdot 2H_2O$ requires C, 36.2; H, 2.8; N, 24.7 per cent), and an oxalate which decomposes without melting above 128° (found: C, 37.5; H, 3.5. $C_8H_6O_4N_4 \cdot 2H_2O$ requires C, 37.2; H, 3.9 per cent).

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¹ *Ber.*, 39, 250 (1906).

Chemical Assay of Streptomycin B (Mannosido - Streptomycin)

MORRIS¹ has recently described the use of a new reagent (0.2 per cent anthrone, a reduction product of anthraquinone, in 95 per cent sulphuric acid) for quantitative determination of carbohydrates. We have found that it can be used not only for distinguishing streptomycin B (a mannoside) from streptomycin A, but also for estimating the amount of the former present in a mixture. The glucosamine moiety, present in both molecules, apparently does not react with the reagent. Results obtained are in accord with those calculated from biological and chemical assays, making the accepted assumption about the relative biological activities of the two streptomycins.

It is hoped to publish elsewhere details of the analytical procedure, with some typical results.

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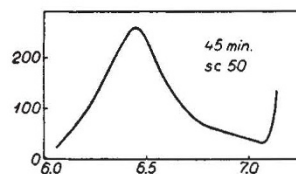
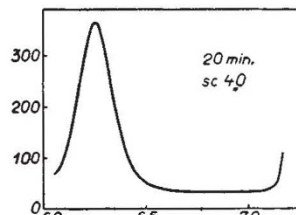
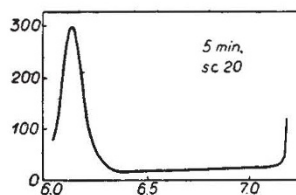
¹ Morris, *Science*, 107, 254 (1948).

Molecular Weight of Malt-Amylase

In an earlier investigation¹, the molecular weight of α -amylase from pig's pancreas^{2,3} was found to be 45,000, as calculated from the values⁴ $s_{20} = 4.50 S$, $D = 8.05 \times 10^{-7}$ c.g.s. and $V_{20} = 0.70$. At this Institute a new method for the isolation of amylase from malt has been developed⁵. It was shown that by repeated purification of the albumin fraction from malt a preparation was obtained with high α -amylase and β -amylase activity. Therefore it was assumed that the amylase activity of malt is localized in the albumin fraction. By ultracentrifugation of the albumin fraction, only one peak was obtained. It must be stated here that malt-amylase is more polydisperse than α -amylase from pancreas.

A determination of the molecular weight was carried out on an amylase preparation with a saccharification activity⁶ of 45,000 and a dextrinizing activity⁷ of 57,400, calculated for the dry substance. The determination was carried out in a buffer solution with a concentration of 0.2 M sodium chloride, 0.03 M primary sodium phosphate and 0.02 M secondary sodium phosphate, pH 7.

Ultracentrifuge measurement. The accompanying sedimentation diagrams were obtained by centrifuging at 65,000 r.p.m.



Sedimentation curves for malt-amylase. In Table 1 the sedimentation constants for different concentrations are shown

Conc. of amylase (per cent)	$s_{20}S$
0.19	4.62
0.28	4.44
0.31	4.96
0.32	4.22
0.41	4.76
0.45	4.12
0.75	4.69
1.00	4.44
1.05	4.39

The sedimentation constant is evidently independent of the concentration. An average value of the sedimentation constant is $s_{20} = 4.52 S$.

Diffusion measurement. The determination of the diffusion constant was carried out in exactly the same way as for α -amylase from pancreas¹. The resulting values of the diffusion constant are shown in Table 2.

Diffusion time (sec.)	34,440	79,020	99,960	129,420	Average value
D_m	6.96	6.70	6.70	6.59	6.74
D_A	6.42	6.20	6.29	6.36	6.32

The agreement between D_m and D_A is good. The average value of $D = 6.53 \times 10^{-7}$ c.g.s. is used in the following. The value $V_{20} = 0.69$ was kindly determined by Prof. C. Drucker. Using these values, the molecular weight of malt-amylase is 54,000. Evidently this value is of the same order of magnitude as that for α -amylase from pancreas. It is an interesting fact that the amylase activity is associated with molecules with about the same molecular weight in both the plant and animal kingdom.

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¹ Danielsson, C.-E., *Nature*, 160, 899 (1947).

² Meyer, K. H., Fischer, E. H., and Bernfeld, P., *Experientia*, 2, 362 (1946); *Helv. Chim. Acta*, 30, 64 (1947).

³ Fischer, E. H., "La purification et l'isolement de l' α -amylase de pancréas". Thèse No. 1094 (Genève, 1947).

⁴ Svedberg, T., and Pedersen, K. O., "The Ultracentrifuge" (Oxford, 1940).

⁵ Danielsson, C.-E., and Sandegren, E., *Acta Chem. Scand.*, 1, 917 (1947).

⁶ Windisch, W., and Kolbach, P., *Woch. Brau.*, 42, 139 (1925).

⁷ Ehrnst, L. E., Yakish, G. J., and Olson, W., *Cereal Chem.*, 16, 724 (1939).