NATURE

We wish to thank Miss H. P. Meggs for preparing the drawing for Fig. 1.

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Formation of Polymers in Textile Fibres

Formation of Polymers in Textile Fibres WHEN wool fibres are exposed to the vapour of ethylene sulphide and water at 50° C., polymerization takes place within the fibres, their elastic properties are modified, and felting is prevented during subsequent working in soap solution¹. Similar reactions take place with other monomeric vinylidene compounds, such as methyl meth-acrylate, if the fibres are first impregnated with ammonium persulphate as a catalyst². It would obviously be a convenience, in practice, if polymers could be formed within wool fibres by treatment with aqueous dispersions of the corresponding monomers instead of with the mixed vapours of water and monomer. A solution to the problem has been found in the use of immobile catalysts, such as iron (Fe++), in conjunction with hydrogen peroxide⁸. Patterns of an all-wool flannel, each 2·5 gm. alr-dry, were immersed in 0·2 per cent ferrous sulphate solution, squeezed, dried and then treated with a mixture of water (200 c.c.) and either methyl meth-acrylate (10 c.c.) or acrylonitrile (10 c.c.) or ne or two hours at 95°-100° C. in presence and absence of hydrogen peroxide (0·03) per cent). For reference purposes, one pattern was treated with methyl methacrylate and water in absence of both iron and hydrogen per-oxide. After treatment the patterns were washed well in running water, conditioned, and re-weighed to determine the gain in weight due to deposition of polymer. The patterns were then hand-milled for 15 minutes in 5 per cent soap solution and the resulting shrinkages are given in the table.

Monomer	Hydro- gen per- oxide	Ferrous sulphate	Time of treat- ment (hours)	Increase in weight (%)	Area shrink- age (%)
None (untreated) Methyl		-	-	-	50.5
methacrylate		+	1	20	24.5
>> >>	-	+	2	45	0.5
** **	-	-	2	12	
27 27	+	+	1	240	
Acrylonitrile	-	+	1	0	48.0
"	+	+	1	32	15.5

Iron alone appears to promote polymerization of methyl meth-acrylate, although the reaction is very much more rapid in presence of hydrogen peroxide, which is essential to the polymerization of acrylonitrile under the above conditions. The unshrinkability con-ferred on the flannel by the above treatments seems to be due to the formation and deposition of polymer within the fibres is unavoidable, especially when large amounts are formed. Reactions such as the pre-ceding are important not merely in providing a simple method of making wool fabrics unshrinkable by means of polymers, but as ex-amples of a general method of modifying the properties of textile fibres. fibres.

One of us (M. L.) is indebted to the International Wool Secretariat and the Australian Wool Board for grants which enabled him to take part in the investigation.

Textile Chemistry Laboratory, University, Leeds. March 2.

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A Bent Pebble

A Bent Pebble DR. JOSEPH NEEDHAR, of the Sino-British Science Co-operation Office, directed my attention to an experiment by Lord Rayleighs on a piece of crown glass under long-continued strain¹. Lord Rayleigh's results suggest further consideration of a curiously curved, small pebble of a fine-grained quartzose sandstone briefly described else-where. Figs. 1 and 2 illustrate almost diametrically opposite aspects of the pebble. By turning the specimen downward from the position shown in Fig. 2 around a horizontal axis through an angle of about 35°, the view represented by Fig. 3 is obtained. These illustrations show that the pebble is equally well-rounded on all sides, characteristic of all water-worn pebbles of the lenticular shape which it would have assumed had it not been folded across the middle. A slight, abnormal inflation is noticeable in Fig. 1 on the concave side along the middle



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the terrain. These circumstances, together with the evidence yielded by the pebble itself, seem to remove any doubt that the pebble was in some way strained by a load due to the glacier. A planed and polished area is seen in Fig. 1 on the outer face of that portion of the pebble which is bent over to the left. This polished area is marked by a few faint strize running perpendicular to the edge on the right produced by bending. While it is admittedly unsound to attach too much sig-nificance to these details, they suggest that deformation was due to holding fast half of the pebble, as might happen by Janming in a crack of the bed-rock or in the glacier fully charged with rock debris, and leaving the other half subject to a forward thrust of the moving ice. J S LEE.

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1 Nature, 145, 29 (1940).

Ultrasonic Velocities in Rock Sections

Ultrasonic Velocities in Rock Sections DURING the course of an investigation relating to the elastic properties of different types of sedimentary and igneous rocks employ-ing the ultrasonic wedge method recently developed by Bhagavantam and Bhimasenachar', I have obtained some results which are likely to be of great importance from two or three different aspects. A preliminary report is accordingly made here. Plates perpendicular to three mutually orthogonal directions are cut from each rock under investigation. Velocities of longitudinal ultrasonic waves with frequencies ranging from I to 10 megacycles along the thickness directions are determined. The results for a few selected samples are given in the accompanying table. Column 2 contains the velocity in a plate parallel to the bedding plane in the case of sedimentary rocks and to that of easy breakage in the case of others. Columns 3 and 4 refer to plates normal to two chosen mutually perpendicular directions in the above planes. Bock (2) (3) (4)

Rock	(2)	(3)	(4)
Bhima limestone	6.20 km./sec.	5.14 km./sec.	5.11 km./sec.
Jubbalpore marble	6.46	5.57	5.57
Gneiss	6.29	5.48	4.32
Pink Hyderabad			
granite	6.62	6.53	4.80
Diabase	7.84	6.91	4.58
Dunite	8.27	7.55	
Deccan trap	8.61	6.97	
Eclogite	7.84	7.84	7.78