

## ORIGINAL ARTICLE

# Bio-inspired porous antenna-like nanocube/nanowire heterostructure as ultra-sensitive cellular interfaces

Biao Kong<sup>1,2</sup>, Jing Tang<sup>1</sup>, Zhangxiong Wu<sup>2</sup>, Cordelia Selomulya<sup>2</sup>, Huanting Wang<sup>2</sup>, Jing Wei<sup>2</sup>, Yongcheng Wang<sup>1</sup>, Gengfeng Zheng<sup>1</sup> and Dongyuan Zhao<sup>1,2</sup>

In this study, an unconventional antenna-like heterostructure comprised of arrays of nanoporous Prussian blue (PB) nanocube heads/TiO<sub>2</sub> nanowire (NW) arms (PB-TiO<sub>2</sub>) is developed for efficient three-dimensional interfacial sensing of small molecules and cellular activities. Inspired by insect tentacles, which are comprised of both target recognition and signal transduction units, one-dimensional TiO<sub>2</sub> NW arrays are grown, followed by selective growth of nanoporous PB nanocubes on the tips of the NW arrays. Due to their high selectivity and bioaffinity toward cells, long biostability under a cell culture adhesion condition (up to 108 h) is obtained, and with its inherent bio-mimetic enzymatic activity, the obtained nanoporous PB nanocubes (head segment) serve as robust substrates for site-selective cell adhesion and culture, which allows for sensitive detection of H<sub>2</sub>O<sub>2</sub>. Simultaneously, the single-crystalline TiO<sub>2</sub> NWs (arm segment) provide efficient charge transport for electrode substrates. Compared with PB-functionalized planar electrochemical interfaces, the PB-TiO<sub>2</sub> antenna NW biointerfaces exhibit a substantial enhancement in electrocatalytic activity and sensitivity for H<sub>2</sub>O<sub>2</sub>, which includes a low detection limit (~20 nM), broad detection range (10<sup>-8</sup> to 10<sup>-5</sup> M), short response time (~5 s) and long-term biocatalytic activity (up to 6 months). The direct cultivation of HeLa cells is demonstrated on the PB-TiO<sub>2</sub> antenna NW arrays, which are capable of sensitive electrochemical recording of cellular activity in real time, where the results suggest the uniqueness of the biomimic PB-TiO<sub>2</sub> antenna NWs for efficient cellular interfacing and molecular recognition.

*NPG Asia Materials* (2014) 6, e117; doi:10.1038/am.2014.56; published online 1 August 2014

## INTRODUCTION

Tentacles, which are the long, functional antennas of insects, provide unique and sensitive signal detection capabilities, such as olfaction, gestation and touching, which can guide the motion and functions of insects.<sup>1</sup> The tip of a tentacle, known as the flagellum, is covered with ultra-sensitive olfaction sensing units that can detect specific molecular targets at extremely low concentrations (as low as several molecules per liter). The signal is then transferred through the trunk segment of the tentacle to the neuron networks.<sup>2</sup> This dual-functional antenna structure of the insect tentacles demonstrates a unique modality for signal capture and transduction and could inspire new electrochemical sensor designs for enhanced sensitivity and performance.<sup>3</sup> In particular, a heterostructure comprised of an ultra-sensitive redox unit and a rapid charge transport unit could substantially expand the capabilities of positioning and targeting of cell patterning and in detecting fluctuations and functions of target molecules at an unprecedented level.

Numerous cellular functions are closely correlated with small molecules, such as O<sub>2</sub>,<sup>4</sup> glucose,<sup>5</sup> glutathione,<sup>6</sup> NO,<sup>7</sup> CO<sup>8</sup> and H<sub>2</sub>O<sub>2</sub>,<sup>9</sup> for enzymatic co-factors, signaling and expression. Among them, H<sub>2</sub>O<sub>2</sub> is an important reactive oxygen species that can diffuse

across cell membranes and lead to oxidative protein modification.<sup>10</sup> Prussian blue (PB) is known as an artificial enzyme peroxidase with porous frameworks, which can efficiently reduce H<sub>2</sub>O<sub>2</sub> at extremely rapid catalytic rates with low overpotentials.<sup>11</sup> For example, molecularly imprinted PB films<sup>12</sup> and PB/carbon nanotubes<sup>13</sup> have been used for biosensing because their low overpotentials can exclude interferences from coexisting substances, such as ascorbic acid (AA), uric acid and glucose. However, previous work predominantly used PB nanocrystals that had been post-modified on flat substrates, and thus, only a limited number of crystal faces were exposed to electrolytes, which resulted in lower electrochemical responses. In addition, a relatively large background current still existed due to the secondary electrochemical reactions that occurred on the planar electrode surface.<sup>14</sup> Although bio-mimetic enzyme-based electrical sensors are attractive tools, their use on a robust lab-on-a-chip platform with fast, sensitive and stable three-dimensional (3D) porous interfaces for localized real-time, long-term monitoring of cellular activities and physicochemical changes have not been demonstrated.<sup>15-19</sup>

Inspired by insect tentacles, herein, we developed an unconventional antenna-like heterostructure based on direct growth of porous

<sup>1</sup>Laboratory of Advanced Materials, Department of Chemistry, Shanghai Key Laboratory of Molecular Catalysis and Innovative Materials, Fudan University, Shanghai, China and

<sup>2</sup>Department of Chemical Engineering, Monash University, Clayton, Victoria, Australia

Correspondence: Professor G Zheng or Professor D Zhao, Laboratory of Advanced Materials, Department of Chemistry, Shanghai Key Laboratory of Molecular Catalysis and Innovative Materials, Fudan University, Shanghai 200433, China.

E-mail: gzheng@fudan.edu.cn or dyzhao@fudan.edu.cn

Received 5 January 2014; revised 5 April 2014; accepted 18 May 2014

PB nanocube heads on the tips of TiO<sub>2</sub> nanowire (NW) arms, which can be used for 3D interface recognition and biosensing. Single-crystalline TiO<sub>2</sub> NWs are first hydrothermally grown on conducting substrates, followed by the direct growth of nanoporous PB nanocubes on the tips of the TiO<sub>2</sub> NW arrays via an etching and seed-assisted process (Figure 1, Supplementary Figure S1). The PB nanocubes (head segment) provide site-selective cell adhesion and growth, a large surface area for catalysis and a high sensitivity and selectivity for H<sub>2</sub>O<sub>2</sub>. The semiconductor TiO<sub>2</sub> NW arms offer fast charge transport from the PB redox centers to the underlying electrode surface due to the excellent electron mobility of TiO<sub>2</sub>, the continuous one-dimensional charge transport channels<sup>20,21</sup> and the low background electrical current noise. As a proof-of-concept design, the bio-mimetic PB/TiO<sub>2</sub> antenna NW heterostructures exhibit high selectivity and bioaffinity toward living cells and excellent biostability under cell culture adhesion conditions (up to 108 h). Compared with that of PB-functionalized planar electrochemical interfaces, the electrocatalytic activity on H<sub>2</sub>O<sub>2</sub> produced by HeLa cells is enhanced by ~550 times on the PB-TiO<sub>2</sub> antenna NWs, which exhibits a high sensitivity of ~20 nM, broad detection range from 10<sup>-8</sup> to 10<sup>-5</sup> M and a fast response time that is within 5 s.

## MATERIALS AND METHODS

### Materials

AA, dopamine, uric acid and Phorbol 12-myristate-3-acetate (PMA) were purchased from Sigma Chemical Co. (St Louis, MO, USA). Tetrabutyl titanate, hydrochloric acid (37 wt%), K<sub>4</sub>Fe(CN)<sub>6</sub> · 3H<sub>2</sub>O, K<sub>3</sub>[Fe(CN)<sub>6</sub>] · 3H<sub>2</sub>O, NaNO<sub>2</sub>, Na<sub>2</sub>SO<sub>3</sub> and H<sub>2</sub>O<sub>2</sub> were purchased from Sinopharm Chemical Reagent Co., Ltd. (Shanghai, China). Dulbecco's Modified Eagle's medium (high glucose), RPMI-1640, fetal calf serum, penicillin G, streptomycin and trypsinase were purchased from GIBCO BRL (Grand Island, NY, USA). The fluorine-doped tin oxide-coated glass ~14 Ω per square substrates was purchased from Wuhan

Ge-ao Ltd. (Wuhan, China). Other reagents were of analytical grade and used as purchased. All the solutions were prepared by Milli-Q water (Merck KGaA, Darmstadt, Germany) and deaerated with high-purity nitrogen before experiments.

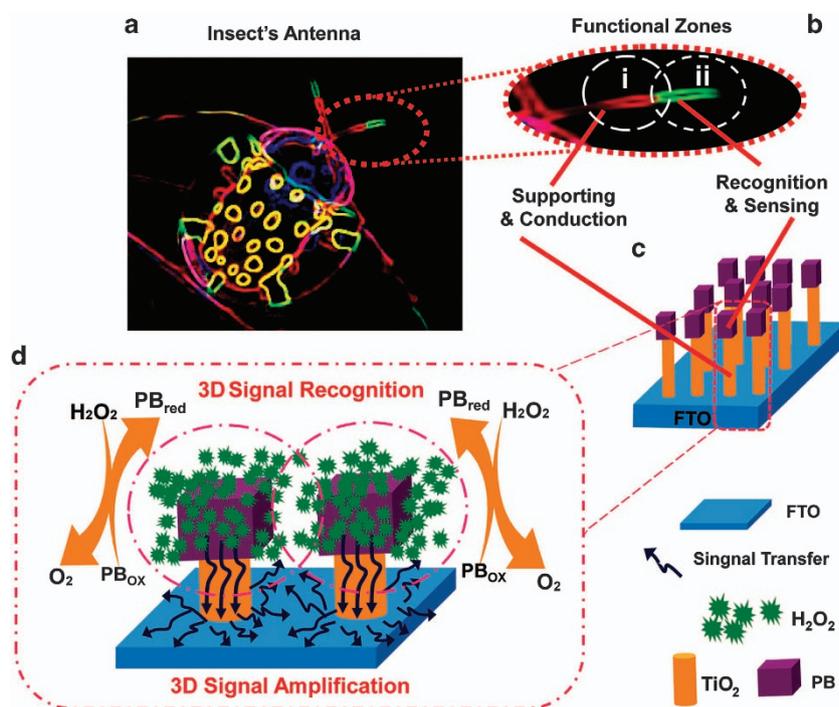
### Methods

The PB-TiO<sub>2</sub> NWs were prepared by a two-step interfacial growth method. TiO<sub>2</sub> NWs were first grown on the fluorine-doped tin oxide-coated glass (~14 Ω per square substrates, Wuhan Ge-ao Ltd.) by a hydrothermal method, as described previously.<sup>22,23</sup> Then, the TiO<sub>2</sub> NWs glass substrates were placed within a 125-ml glass bottle, containing 80 ml of HCl (0.005 M) and 136 mg of K<sub>3</sub>[Fe(CN)<sub>6</sub>] · 3H<sub>2</sub>O, and stirred for 30 min. The bottle was placed in an oven at 85 °C for 24 h to form the PB nanocrystal seeds. The TiO<sub>2</sub> NWs with PB seeds were washed with neat ethanol and water thoroughly to remove the adsorbed nanocrystals on the TiO<sub>2</sub> NWs surface. To form the PB hierarchical structure on TiO<sub>2</sub> NWs, K<sub>4</sub>Fe(CN)<sub>6</sub> · 3H<sub>2</sub>O (125 mg) was first added to a HCl solution (0.05 M, 80 ml) under stirring for 30 min. Then, the TiO<sub>2</sub>-PB seeds substrate was slowly immersed into the above mixture. After that, the container was placed into an oven and heated at 85 °C for 6–18 h. The obtained PB-armed TiO<sub>2</sub> NWs was taken from the container, washed with deionized water and dried in a vacuum oven at 70 °C for 12 h.

## RESULTS AND DISCUSSION

### Fabrication of PB-TiO<sub>2</sub> antenna NW heterostructures

The controlled synthesis of the PB-TiO<sub>2</sub> antenna NW heterostructures is performed in two steps. One-dimensional TiO<sub>2</sub> NWs are first hydrothermally grown on fluorine-doped tin oxide substrates, followed by a seed-induced growth of PB nanocubes on the tips of the pre-formed TiO<sub>2</sub> NWs (Experimental section in the Supplementary Information). Scanning electron microscopy images show that the diameters and densities of the TiO<sub>2</sub> NW arrays are well tuned by the acidity of the precursor solution.<sup>22,23</sup> Increasing the solution acidity

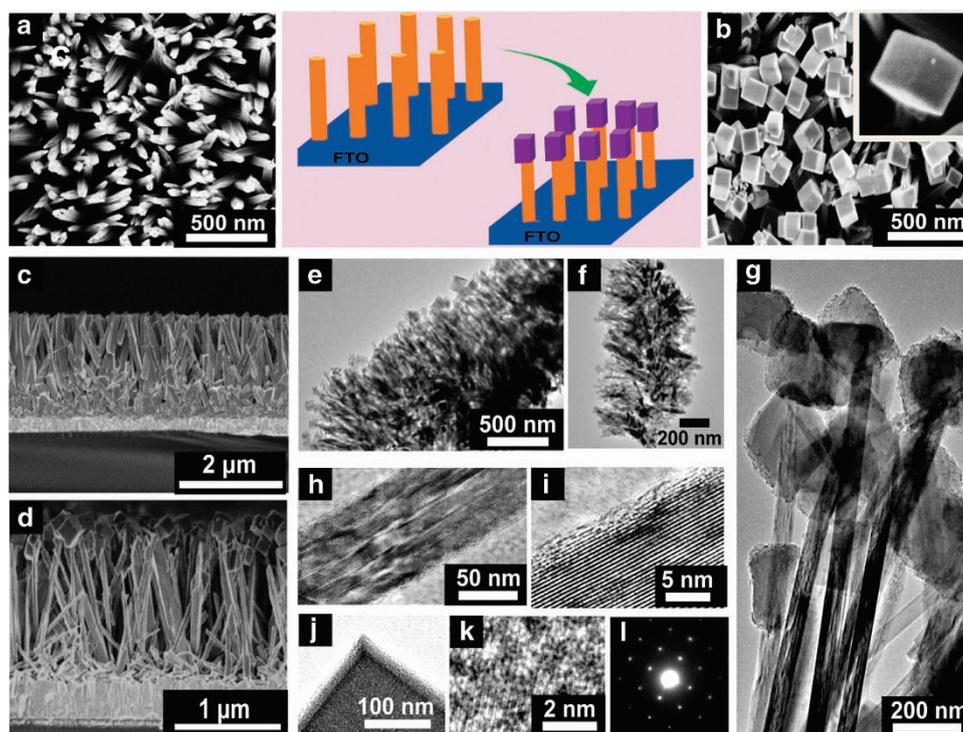


**Figure 1** Schematic illustration of three-dimensional (3D) signal recognition and amplification based on porous bio-mimetic antenna nanowire arrays. (a) A typical drawing of an insect's antennas according to their different functional zones. (b) Enlarged drawing of an antenna with (i) supporting and conduction and (ii) recognition and sensing zones. (c) Designed nanoporous bio-mimetic antenna arrays growth on a conducting substrate. (d) 3D signal recognition and amplification based on the nanoporous bio-mimetic antenna arrays.

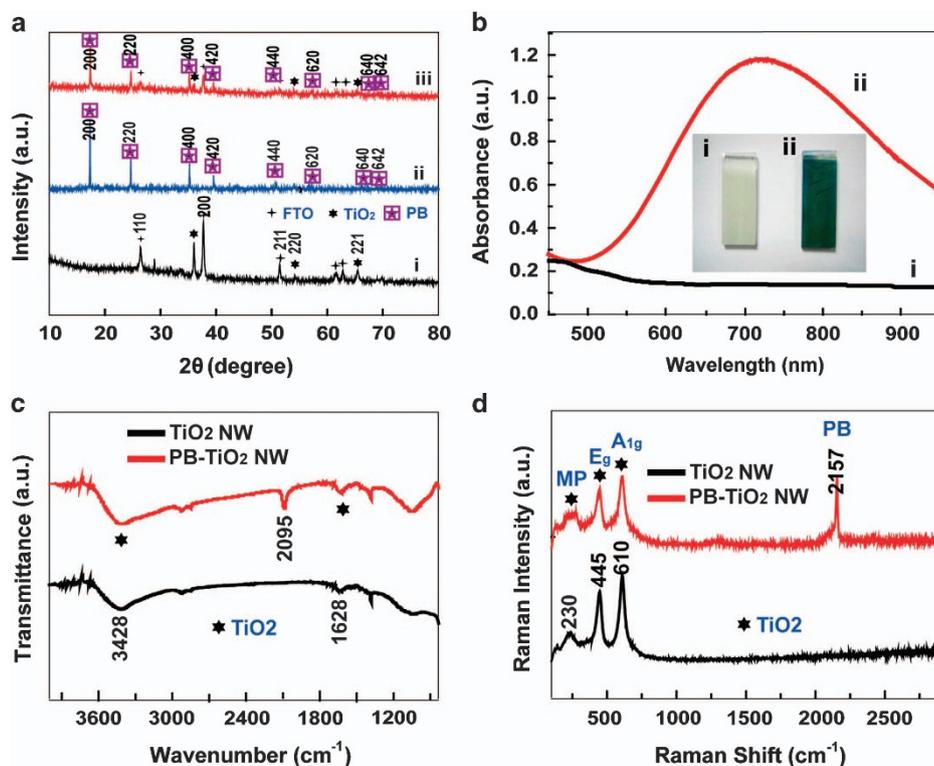
leads to a decrease in both diameter and density (Figure 2a, Supplementary Figures S2a and c). Subsequently, the tips of the TiO<sub>2</sub> NWs are etched using diluted HCl, followed by hydrothermal growth of PB nanocubes (Supplementary Figure S1).<sup>24</sup> At low TiO<sub>2</sub> NW densities, each NW tip is covered with a cluster of nanocubes, which substantially exceed the diameter of the corresponding NW arms (Supplementary Figure S2b). At intermediate NW densities, the measured average diameter and interspacing between adjacent NWs are ~100 and 150 nm, respectively (Figure 2a). The nanocubes with well-defined shapes are grown from each TiO<sub>2</sub> NW (Figure 2b, Supplementary Figure S3). Cross-sectional scanning electron microscopy images also clearly show the bio-mimetic antenna-like heterostructure formation (Figures 2c and d). Compared with bare TiO<sub>2</sub> NWs, the NW tips are covered by a thin layer of PB nanocubes. The edge length of the nanocubes is calculated to be ~150 nm, which is similar to the average interspacing between adjacent TiO<sub>2</sub> NWs. The top-view and side-view transmission electron microscopy (TEM) images show that the number of nanocubes and underlying TiO<sub>2</sub> NWs are similar, which suggests that essentially, each NW tip serves as a growth base for individual nanocubes (Figures 2e and f). This structure provides sufficient space surrounding the crystallized PB nanocubes while maintaining a close contact between the PB nanocubes with the TiO<sub>2</sub> NW bases. When the diameter and density of the TiO<sub>2</sub> NWs are further increased, the interspacing between adjacent NWs is less than the average NW diameter. Thus, the nanocubes on the NW tips accumulate, which forms a multi-layered nanocube assembly (Supplementary Figure S2d). The heterostructure of the PB nanocube head/TiO<sub>2</sub> NW arm is revealed by TEM

(Figure 2g, Supplementary Figure S4). The single TiO<sub>2</sub> NWs (Figures 2h and i) and single PB nanocubes (Figures 2j and k) in the heterostructures are further characterized by high-resolution TEM (HRTEM). Both the HRTEM and selected area electron diffraction pattern (Figure 2l) demonstrate the crystallized characteristics of the TiO<sub>2</sub> NWs and PB nanocrystals.

The crystal structure of the PB-TiO<sub>2</sub> antenna NW heterostructures is further examined by X-ray diffraction, which shows a combination of the diffraction peaks of TiO<sub>2</sub> and PB (Figure 3a). No additional peaks other than those of fluorine-doped tin oxide substrates are observed, which indicates the high purity of the samples. Compared with pristine TiO<sub>2</sub> NWs that have negligible absorption in the visible range, the ultraviolet–visible spectra of the PB-TiO<sub>2</sub> antenna NWs show a substantial absorption increase in the range of 500–900 nm with a maximum at 714 nm, which corresponds to the inter-metal charge transfer bands from Fe<sup>2+</sup> to Fe<sup>3+</sup> in PB nanocrystals (Figure 3b).<sup>25</sup> The corresponding energy dispersive X-ray analysis of the PB-TiO<sub>2</sub> NWs shows clear Fe and Ti signals (Supplementary Figure S5). The growth of PB-TiO<sub>2</sub> antenna NWs is further demonstrated by the infrared spectra. In addition to the infrared absorption of pristine TiO<sub>2</sub> NWs centered at 3428 and 1628 cm<sup>-1</sup>, a new absorption band at 2095 cm<sup>-1</sup> is observed for the PB-TiO<sub>2</sub> antenna NWs, which corresponds to the C≡N stretching mode in the Fe<sup>II</sup>(C≡N)/Fe<sup>III</sup> pair of PB nanocrystals (Figure 3c).<sup>26</sup> Furthermore, the Raman spectrum of the PB-TiO<sub>2</sub> antenna NWs shows the characteristic band of PB at 2157 cm<sup>-1</sup>, which further confirms the successful incorporation of PB over TiO<sub>2</sub> NWs (Figure 3d).<sup>13</sup>



**Figure 2** Synthesis and characterization of bio-mimetic antenna Prussian blue (PB)-TiO<sub>2</sub> heterostructure arrays. (a) Top-view scanning electron microscopy (SEM) images of the TiO<sub>2</sub> nanowire (NW) arrays on fluorine-doped tin oxide (FTO)-coated glass substrates at an intermediate density. (b) Top-view SEM images of PB-TiO<sub>2</sub> NWs at an intermediate density via an etching and seed growth method. (c, d) Side-view SEM images of (k) TiO<sub>2</sub> NW arrays and (l) PB-TiO<sub>2</sub> NW arrays at an intermediate density. (e) Side-view, (f) top-view and (g) enlarged transmission electron microscopy (TEM) images of synthesized PB-TiO<sub>2</sub> NW arrays at an intermediate density, as in e. (h, i) High-resolution TEM (HRTEM) and enlarged HRTEM image of a single TiO<sub>2</sub> NW. (j, k) HRTEM and enlarged HRTEM image of a single PB nanocube. (l) Selected area electron diffraction (SAED) pattern of a single PB nanocube.



**Figure 3** (a) X-ray diffraction spectrum of TiO<sub>2</sub> nanowires (NWs), Prussian blue (PB) nanocrystals and PB-TiO<sub>2</sub> NWs. The peaks of PB, blank TiO<sub>2</sub> and fluorine-doped tin oxide substrates are marked. (b) The ultraviolet–visible diffused reflectance spectra of blank TiO<sub>2</sub> NWs and PB-TiO<sub>2</sub> NWs. Inset: the corresponding images of blank TiO<sub>2</sub> NWs and PB-TiO<sub>2</sub> NWs. (c) Fourier transform infrared spectroscopy spectra of the TiO<sub>2</sub> NWs (black curve) and PB-TiO<sub>2</sub> NWs (red curve). (d) Raman shift spectra of TiO<sub>2</sub> NWs (black curve) and PB-TiO<sub>2</sub> NWs (red curve).

The synthesis mechanism of the proposed etching and seed-assisted process is as follows. First, the intermediate PB seed-modified TiO<sub>2</sub> NWs are observed on the tips of pristine TiO<sub>2</sub> NWs, in which the composite TiO<sub>2</sub> NWs have identical close-packed cubic structures (Supplementary Figure S6). Thus, no clear phase separation occurs on the TiO<sub>2</sub> NW interfaces during the hydrothermal growth process. The removal of PB nanocrystals by an alkaline solution at room temperature for 5 h from the PB-TiO<sub>2</sub> NWs increased the porous property on the top of the TiO<sub>2</sub> NWs (Supplementary Figure S7), which suggests the existence of acid etching during the hydrothermal process. For comparison, the growth without PB seeds does not result in any nanocube structures on the tips of the TiO<sub>2</sub> NWs (Supplementary Figure S8). However, the growth with acid etching only and in the absence of seeds shows that no nanocube clusters formed on the TiO<sub>2</sub> NWs during the hydrothermal process (Supplementary Figure S9); only sporadic PB nanocrystals are adsorbed on the surface of the TiO<sub>2</sub> NWs.

### Electrochemical performance

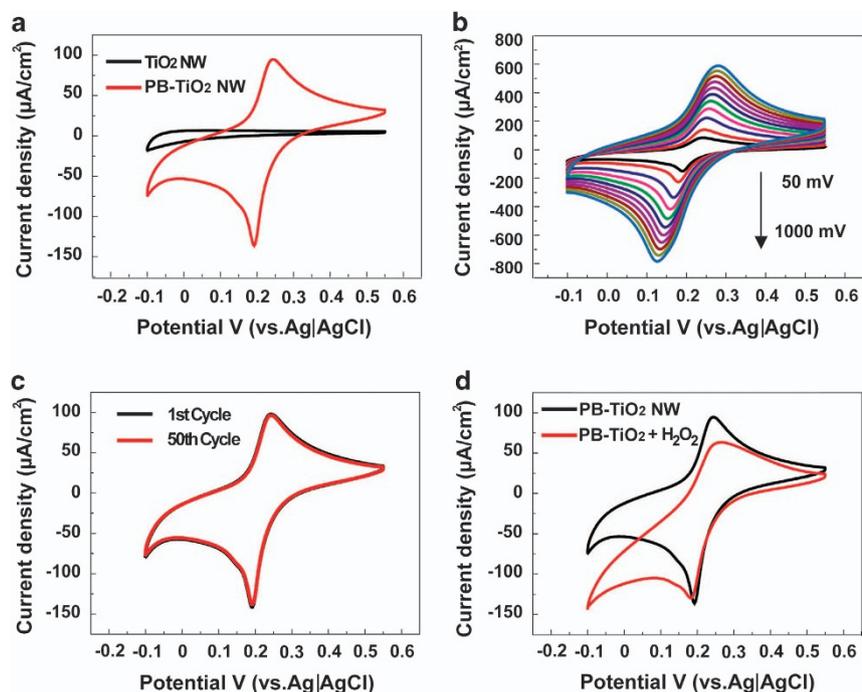
The electrochemical properties of the PB-TiO<sub>2</sub> antenna NW heterostructure are investigated by the cyclic voltammetry method. For comparison, the pristine TiO<sub>2</sub> NWs are also measured under similar conditions. No redox peaks except for the capacitive current are observed for the pristine TiO<sub>2</sub> NW electrode, whereas the PB-TiO<sub>2</sub> antenna NW electrode has a pair of redox peaks at 0.23 and 0.18 V (Figure 4a), which corresponds to the reversible conversion of PB to Prussian white.<sup>25</sup> The scan rate dependency for the voltammetry profile of the PB-TiO<sub>2</sub> antenna NW electrode in the range 50–1000 mV s<sup>-1</sup> is then presented (Figure 4b), in which the electrochemical stability of the PB-TiO<sub>2</sub> antenna NWs is

demonstrated by repeating cyclic voltammetry measurements at a scan rate of 50 mV s<sup>-1</sup>. There is no observable difference in either current level or peak positions for the cyclic voltammetry curves after 50 cycles (Figure 4c), which confirms the stable structure of the immobilized PB nanocubes on the TiO<sub>2</sub> NW tips.

The capability of using PB-TiO<sub>2</sub> antenna NWs as an amperometric sensing modality for H<sub>2</sub>O<sub>2</sub> is further investigated because it is well-known that the reduced form of PB exhibits a high catalytic activity for H<sub>2</sub>O<sub>2</sub> reduction.<sup>27</sup> During this process, PB acts as an electron transport mediator between the electrode and H<sub>2</sub>O<sub>2</sub> in a solution. The presence of H<sub>2</sub>O<sub>2</sub> (5 mM in phosphate buffer saline (PBS), pH of 6.0) leads to a clear increase in current density (corresponding to H<sub>2</sub>O<sub>2</sub> reduction) at a lower overpotential (that is, -50 mV versus Ag/AgCl) (Figure 4d). The calculated electron transfer rate constant,  $k_s$ , is  $5.72 \pm 0.05 \text{ s}^{-1}$ , which is much greater than that for most of the H<sub>2</sub>O<sub>2</sub> electrodes reported previously, such as the ordered nanoporous niobium oxide film ( $0.28 \text{ s}^{-1}$ ),<sup>28</sup> NaY zeolite/glassy carbon electrode ( $0.78 \pm 0.04 \text{ s}^{-1}$ )<sup>29</sup> and colloidal Au/carbon paste ( $1.21 \pm 0.08 \text{ s}^{-1}$ ).<sup>30</sup>

### PB-TiO<sub>2</sub> antenna NW arrays for H<sub>2</sub>O<sub>2</sub> sensing

The sensor performance of PB-TiO<sub>2</sub> antenna NWs in detecting H<sub>2</sub>O<sub>2</sub> is extensively investigated. Because coexisting molecular interferences, such as AA, O<sub>2</sub> and so on, may affect the electrochemical determination of H<sub>2</sub>O<sub>2</sub>, the bias potential should be prudently selected to optimize the cathodic current and sensitivity obtained at the PB-TiO<sub>2</sub> antenna NW electrodes. Amperometric experiments were performed to investigate the responses of PB-TiO<sub>2</sub> antennas from H<sub>2</sub>O<sub>2</sub> at various potentials of -0.10, -0.05, 0.00, 0.05 and 0.10 V (versus Ag/AgCl). Several interference molecules, which included O<sub>2</sub>, Na<sub>2</sub>SO<sub>3</sub>, uric acid, 3,4-dihydroxyphenylacetic acid, NaNO<sub>2</sub> and AA, were tested



**Figure 4** Electrochemical performance of bio-mimetic Prussian blue (PB)-TiO<sub>2</sub> antenna arrays. (a) Cyclic voltammograms (CVs) of the blank TiO<sub>2</sub> nanowires (NWs) (black curve) and PB-TiO<sub>2</sub> NWs (red curve) in N<sub>2</sub>-saturated 0.05-M phosphate buffer saline (PBS) solution (pH of 6.0) at a scan rate of 50 mVs<sup>-1</sup>. An Ag/AgCl electrode was used as a reference electrode. (b) CVs of PB-TiO<sub>2</sub> NWs in N<sub>2</sub>-saturated 0.05-M PBS solution (pH of 6.0) at different scan rates: 50–1000 mVs<sup>-1</sup> from the inside to the outside. (c) The CVs of the first cycle (black curve) and 50th cycle (red curve) of the PB-TiO<sub>2</sub> NWs in the N<sub>2</sub>-saturated 0.05-M PBS solution (pH of 6.0) at a scan rate of 50 mVs<sup>-1</sup>. (d) The CVs of a PB-TiO<sub>2</sub> NWs electrode in the N<sub>2</sub>-saturated 0.05-M PBS solution (pH of 6.0) in the absence (black curve) and presence of 5 mM H<sub>2</sub>O<sub>2</sub> (red curve) at a scan rate of 50 mVs<sup>-1</sup>.

(Figure 5a). In general, a low anodic current is obtained for the interference molecules at relatively negative potentials. For example, the ratio of anodic current between H<sub>2</sub>O<sub>2</sub> to AA (0.1 mM each) increases from 6.8 to 55 A when the applied potential is reduced from 0.10 to -0.05 V (versus Ag/AgCl), which increases the selectivity.<sup>31</sup> Hence, -0.05 V (versus Ag/AgCl) is selected as the optimized operational bias potential. In contrast, control experiments of pristine TiO<sub>2</sub> NWs and TiO<sub>2</sub> NWs post-modified with PB (TiO<sub>2</sub> NWs + PB) do not show a similar high current signal or signal ratios, even at an applied potential of -0.05 V (Figures 5b and c), which suggests that the direct growth and attachment of PB nanocubes on TiO<sub>2</sub> NW tips enhance the sensitivity and selectivity. In addition, these data confirm the bio-mimetic enzymatic amplification nature of the H<sub>2</sub>O<sub>2</sub> catalysis at the PB-TiO<sub>2</sub> antenna NW electrodes. The long-term stability of PB-TiO<sub>2</sub> antenna NW electrodes is exemplified by repeating cyclic voltammetry cycles at different bias voltages. The PB-TiO<sub>2</sub> antenna maintains 95% of its initial signal responses, even after 1000 cycles (Figure 5d), which is much more sensitive and stable than pristine TiO<sub>2</sub> NWs and TiO<sub>2</sub> NWs post-modified with PB, which is excellent for long-term signal monitoring (Figures 5e and f).

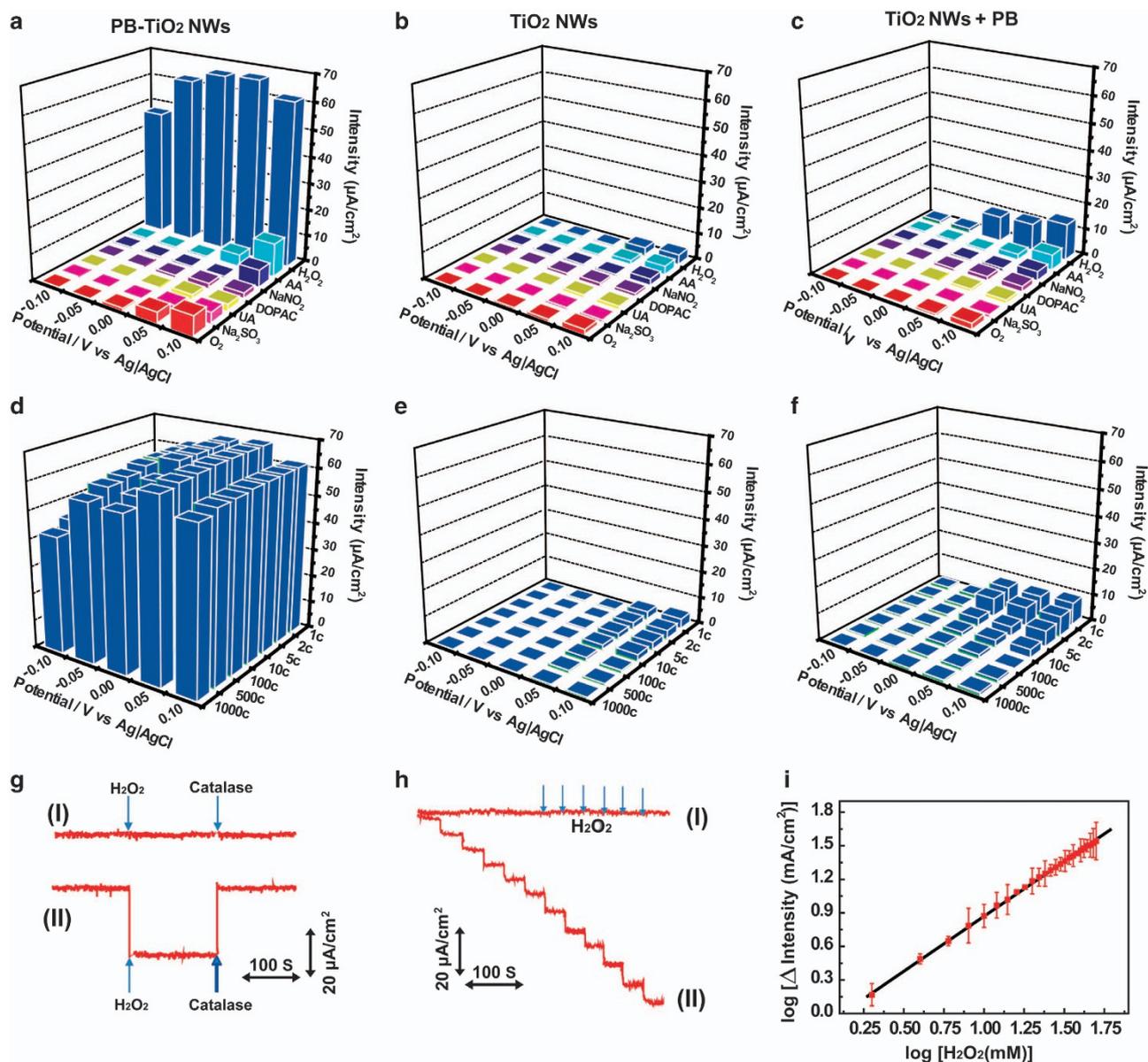
The amperometric measurement is further investigated by determining the electrocatalytic activity of the PB-TiO<sub>2</sub> antenna NWs for H<sub>2</sub>O<sub>2</sub>. A substantial cathodic current change is observed at the PB-TiO<sub>2</sub> antenna NW electrodes at -50 mV when 25 μM of H<sub>2</sub>O<sub>2</sub> is introduced in the buffer, whereas essentially no response is obtained at the pristine TiO<sub>2</sub> NWs under otherwise similar conditions (Figure 5g). The contribution of the observed anodic current to the catalysis of H<sub>2</sub>O<sub>2</sub> is confirmed by adding catalase,<sup>10,32</sup> a selective scavenger of H<sub>2</sub>O<sub>2</sub>, into the solution with H<sub>2</sub>O<sub>2</sub>. A large decrease in the cathodic current to essentially the background level is observed,

which indicates that the signal response can be ascribed to the catalysis of H<sub>2</sub>O<sub>2</sub> by the PB-TiO<sub>2</sub> antenna NW electrodes.

Amperometric responses of the PB-TiO<sub>2</sub> antenna NWs with successive addition of H<sub>2</sub>O<sub>2</sub> are conducted at the optimized potential of -50 mV. The stepwise current signal correlates well with each addition of H<sub>2</sub>O<sub>2</sub>, whereas an almost negligible current response is observed for the pristine TiO<sub>2</sub> NWs (Figure 5h). The calibration plot of the current change obtained at the PB-TiO<sub>2</sub> antenna NWs with the H<sub>2</sub>O<sub>2</sub> concentration is summarized (Figure 5i), and the analytical performance of the PB-TiO<sub>2</sub> antenna NW-based H<sub>2</sub>O<sub>2</sub> biosensor at the optimum conditions is listed in Supplementary Table S1. Other previously reported H<sub>2</sub>O<sub>2</sub> biosensors based on TiO<sub>2</sub> or titanate nanostructures are also listed in Supplementary Table S1 for comparison. The PB-TiO<sub>2</sub> antenna NWs exhibit an excellent sensor performance, which includes their higher selectivity, wider linear detection range and lower detection limit. Specifically, the dynamic linear range of 10<sup>-8</sup> to 10<sup>-5</sup> M to detect H<sub>2</sub>O<sub>2</sub> at the applied potential of -50 mV is significantly wider than those obtained in previous H<sub>2</sub>O<sub>2</sub> biosensing interfaces<sup>33,34</sup> and fulfills the requirement of real-time, long-term tracking of an H<sub>2</sub>O<sub>2</sub> concentration (Figure 5i). Moreover, the PB-TiO<sub>2</sub> antenna NW biosensors can be tested and stored at room temperature and can be repeatedly tested for over 180 days without substantial degradation (Supplementary Figure S10), which suggests the high stability of the PB-TiO<sub>2</sub> antenna NW electrodes.

#### Cell culture and extracellular H<sub>2</sub>O<sub>2</sub> sensing

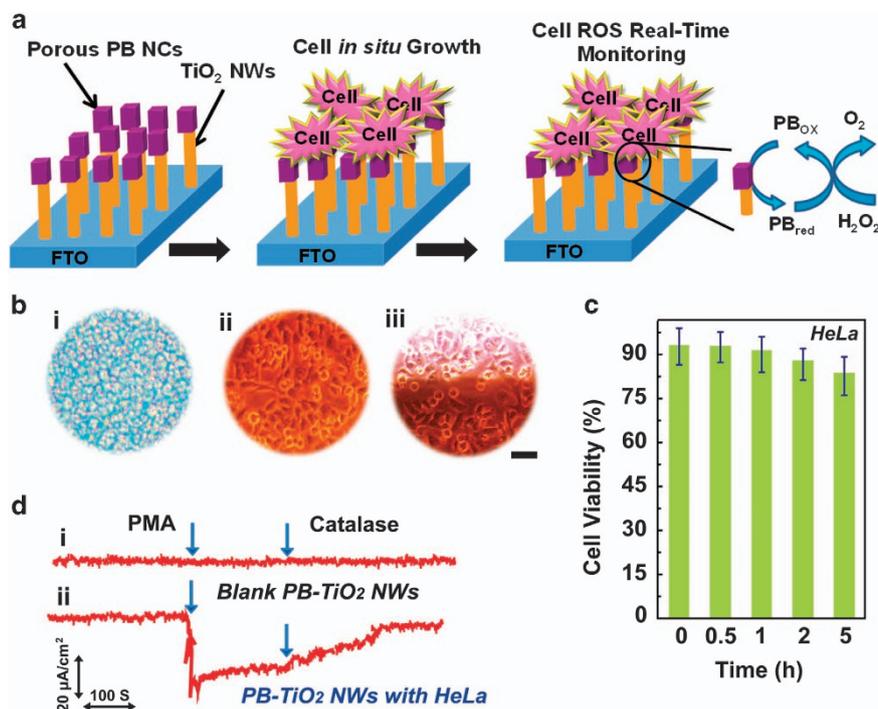
The capability of PB-TiO<sub>2</sub> antenna NW electrodes for direct growth of cells and extracellular H<sub>2</sub>O<sub>2</sub> detection is further demonstrated. The uniform coverage of PB nanocubes offers an excellent substrate for



**Figure 5** The selectivity, sensitivity, stability and detection range of bio-mimetic Prussian blue (PB)-TiO<sub>2</sub> nanowire (NW) antenna arrays. (a–c) The selectivity, sensitivity and (d–f) stability profile of the present PB-TiO<sub>2</sub> NW (a,d) electrodes obtained at different applied potentials:  $-0.10$ ,  $-0.05$ ,  $0.00$ ,  $0.05$  and  $0.10$  V versus Ag/AgCl. The TiO<sub>2</sub> NWs (b,e) and traditional post-modified PB nanocrystals on TiO<sub>2</sub> NW (c,f) electrodes were used as control experiments. (g) Amperometric responses obtained at the (i) blank TiO<sub>2</sub> NWs and (ii) PB-TiO<sub>2</sub> NWs electrode in 50 mM phosphate buffer saline (PBS) (pH of 6.0) containing 100 mM glucose at the applied potential of 0.0 V versus Ag/AgCl with the addition of the final concentration of 25 μM H<sub>2</sub>O<sub>2</sub> and 500 U ml<sup>-1</sup> catalase. (h) Typical amperometric responses of (i) blank TiO<sub>2</sub> NWs and (ii) PB-TiO<sub>2</sub> NWs to successive additions of 10 μM H<sub>2</sub>O<sub>2</sub> at an applied potential of  $-0.05$  V versus Ag/AgCl in 50 mM PBS (pH of 6.0). (i) Log-response curve of steady-state currents obtained at the PB-TiO<sub>2</sub> NW electrode against concentrations of H<sub>2</sub>O<sub>2</sub>.

cell attachment and growth, which can subsequently produce H<sub>2</sub>O<sub>2</sub> via different cell functions (Figure 6a).<sup>35–37</sup> The fluorescence images of HeLa cells grown on top of the PB-TiO<sub>2</sub> antenna NWs clearly confirm a good coverage of cells on the NWs (Figure 6b). The cell viability is measured by [3-(4,5-dimethylthiazol-2-yl)-2,5-diphenyltetrazolium bromide] (MTT) in a Nun Immune Omni Tray (Angle Nun International, Rochester, NY, USA; Experimental Section in Supporting Information, Supplementary Figures S11–S13), which shows that a high HeLa cell viability (>80%) is maintained for growth for up to 5 h (Figure 6c). The measured amperometric responses obtained at the bio-interface between the PB-TiO<sub>2</sub> antenna

NWs and HeLa cells are 50 mM PBS (pH of 6.0) at an applied potential of  $-50$  mV versus Ag/AgCl (Figure 6c). When 50 mM of PMA is injected into the HeLa cell-NW assay, an increase in the cathodic current is observed (Figure 6d). This phenomenon is attributed to the effect of PMA for inducing H<sub>2</sub>O<sub>2</sub> production from the cells.<sup>10</sup> An anodic current increase of  $\sim 32.5$  μA is obtained in 15 s, which is similar to the time scale reported previously.<sup>38</sup> No response is observed at the bare TiO<sub>2</sub> NWs with the same addition of PMA. In addition, the injection of the catalase solution (300 U ml<sup>-1</sup> in PBS) reduces the current level to almost the background current level; catalase is known to inhibit the PMA function.<sup>10</sup> Accordingly,



**Figure 6** Ultra-sensitive bio-interface based on bio-mimetic antenna Prussian blue (PB)-TiO<sub>2</sub> nanowire (NW) arrays. (a) Schematic illustration of the PB-TiO<sub>2</sub> NW arrays bio-interface for *in-situ* living cell culture and real-time three-dimensional recognition and biosensing. (b) Fluorescence microscopic images of (i) PB-TiO<sub>2</sub> NWs, (ii) PB-TiO<sub>2</sub> NWs with *in situ*-cultured HeLa cells, and (iii) the interface of PB-TiO<sub>2</sub> NWs and culture dish with *in situ*-cultured HeLa cells (12 h). The scale bar is 50 μm for (i–iii). (c) Cell viability of cells cultured on PB-TiO<sub>2</sub> NWs with the amount of testing time. (d) Amperometric responses obtained at the (i) blank PB-TiO<sub>2</sub> NWs and (ii) PB-TiO<sub>2</sub> NWs with HeLa cell incubation. The measurement was performed in 50 mM phosphate buffer saline (pH of 6.0) that contained 100 mM glucose at the applied potential of  $-0.05$  V versus Ag/AgCl after the final injection concentrations of 50 mM Phorbol 12-myristate-3-acetate (PMA) and 300 U ml<sup>-1</sup> catalase.

the increase in cathodic current at the PB-TiO<sub>2</sub> antenna NW bio-interface located near the cells is ascribed to the enzymatic reduction of H<sub>2</sub>O<sub>2</sub>, which is effectively mediated by the PB nanocubes grown on the TiO<sub>2</sub> NWs interface.

The much enhanced sensitivity of the bio-mimetic PB-TiO<sub>2</sub> antenna NWs is attributed to the unique bio-inspired head/arm heterostructure, in which intimate contact between the PB nanocube heads and the TiO<sub>2</sub> NW arms can create synergistic properties of both components. The porous PB nanocubes offer a robust substrate for site-selective cell adhesion and cultivation of living cells because the porous nanocubes exhibit high selectivity and bioaffinity toward cells and have excellent biostability under the cell culture adhesion condition (up to 108 h) (Supplementary Figure S14). Furthermore, the porous nanocrystals can also serve as long-term stable and sensitive sensing elements for H<sub>2</sub>O<sub>2</sub> due to their inherent bio-mimetic enzymatic activity, high surface area and 3D stereo space-based signal molecules for touching and recognition. In addition, compared with conventional planar PB-covered electrochemical interfaces,<sup>39,40</sup> the electrocatalytic activity is enhanced at the PB-TiO<sub>2</sub> antenna NW biointerfaces due to the rapid charge transport realized by the one-dimensional TiO<sub>2</sub> NW structure. Thus, this PB-TiO<sub>2</sub> antenna NW demonstrates a new sensing platform to reliably and durably detect extracellular molecules.

## CONCLUSIONS

In summary, a bio-mimetic PB-TiO<sub>2</sub> antenna NW array-based bio-interface is demonstrated for the first time as an excellent structure that can recognize 3D signal molecules and that can be used for

biosensing. Living cell adhesion and *in-situ* cultivation on PB-TiO<sub>2</sub> antennas, which integrates with sensitive real-time monitoring of cellular messenger molecules, are realized by the dual-functional PB nanocube heads and TiO<sub>2</sub> NW arms. The optimized PB-TiO<sub>2</sub> antenna NW biointerfaces exhibit a remarkable sensing performance in detecting H<sub>2</sub>O<sub>2</sub> with a high sensitivity and selectivity, a broad detection range from 10<sup>-8</sup> to 10<sup>-5</sup> M, a low detection limit down to 20 nM and a short response time that is within 5 s. In addition, bio-mimetic PB nanocrystals are stably anchored on the TiO<sub>2</sub> NWs and can maintain long-term biocatalytic activity (up to ~180 days). The capability of the PB-TiO<sub>2</sub> antenna NW biointerfaces in determining extracellular H<sub>2</sub>O<sub>2</sub> released from human tumor cells is further demonstrated. This work not only provides a method for 3D interface recognition and biosensing on the hierarchical bio-mimetic nanostructured semiconductors but also suggests a general approach for durable, reliable biomolecule detection in biological systems.

## CONFLICT OF INTEREST

The authors declare no conflict of interest.

## ACKNOWLEDGEMENTS

We thank the following funding agencies for supporting this work: the National Key Basic Research Program of China (2013CB934104, 2012CB224805), the NSF of China (20890123, 21322311, and 21071033), the Program for New Century Excellent Talents in University (NCET-10-0357), the Program for Professor of Special Appointment (Eastern Scholar) at the Shanghai Institutions of Higher Learning and the Deanship of Scientific Research of the King Saud University (IHCRC#14-102) and the Shanghai Science and Technology Municipality (14JC1490500). BK and JT thank the

Scholarship Award for Excellent Doctoral Student granted by the Ministry of Education of China and the Interdisciplinary Outstanding Doctoral Research Funding of the Fudan University.

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